

A Quantum Gas of Polar Molecules

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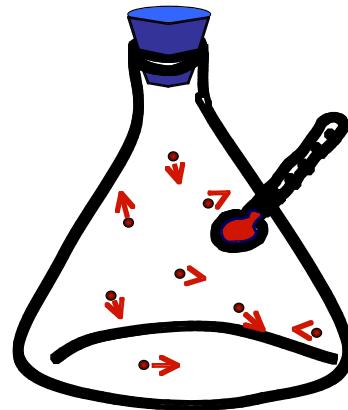
Josh Zirbel
Silke Ospelkaus
Avi Pe'er
Marcio Miranda
Brian Neyenhuis
Dajun Wang



Deborah Jin
Jun Ye
Carl Wieman

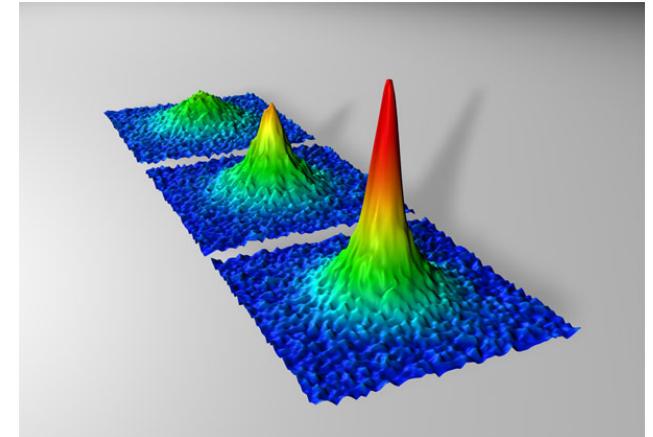
Keck Foundation, NIST, and NSF

Ultracold Atomic Quantum Gases



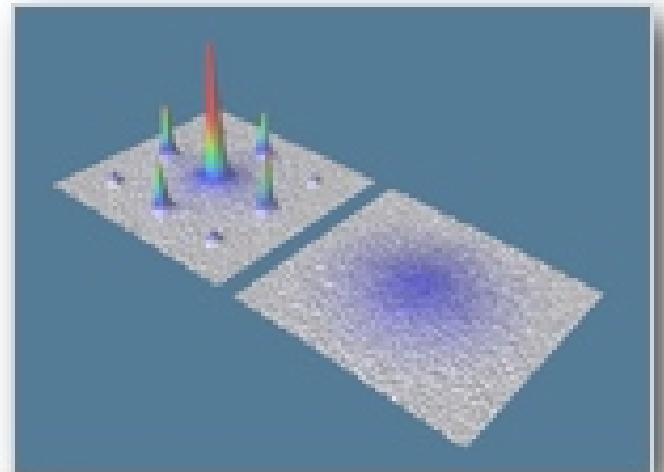
$< 1 \mu\text{K}$

Bose-Einstein Condensation
Degenerate Fermi Gas



strongly correlated systems

- * superfluidity in Fermi gases
- * quantum phase transitions with fermions/bosons in optical lattices
- * and many more...

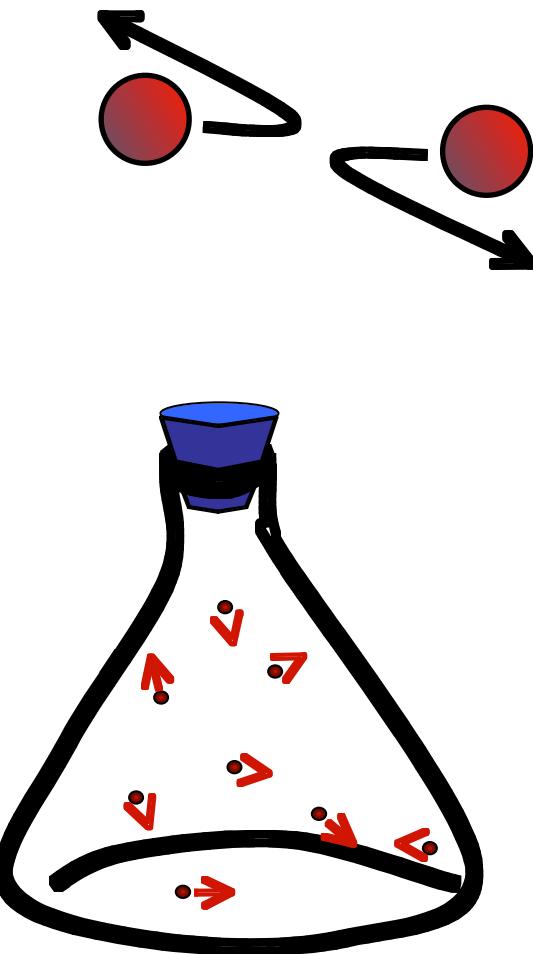


Interactions are Key

Atoms

Contact interactions

- * short range $\delta(R)$
- * isotropic



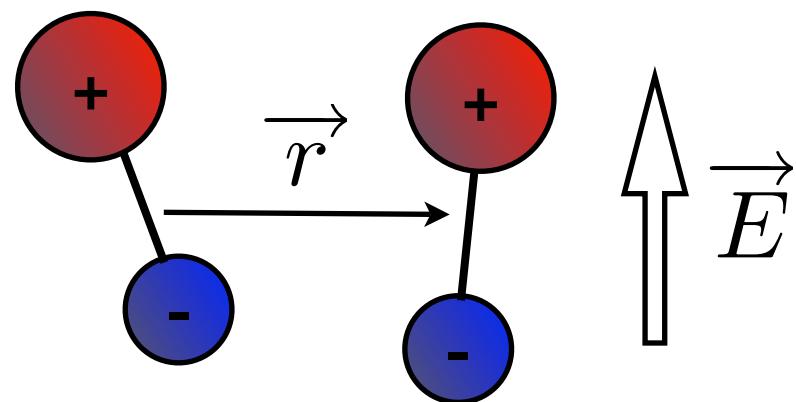
Why Polar Molecules?

dipole-dipole interactions

anisotropic

long-range ($1/r^3$)

tunable dipoles



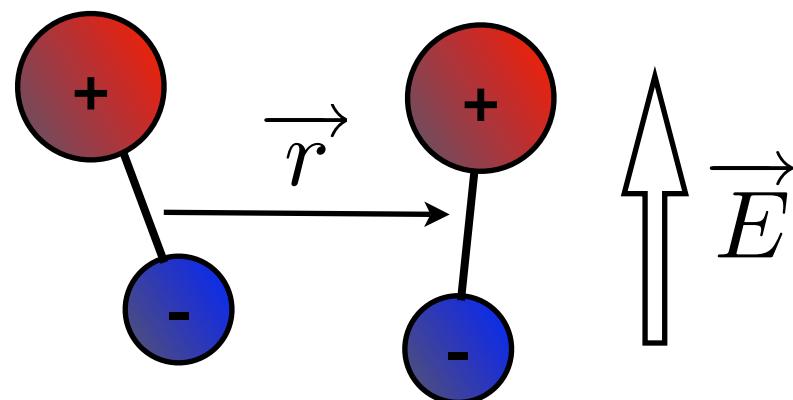
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$$\frac{(\text{Debye})^2}{(\text{Bohr magneton})^2} \cdot c^2 = 10^4$$

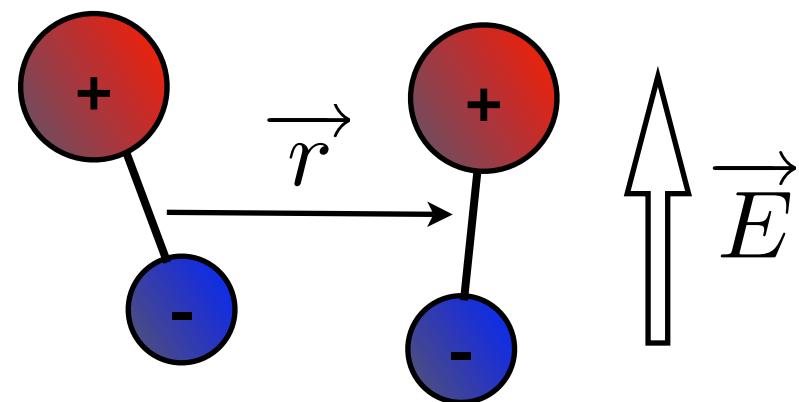
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complex internal structure offers new ways of control

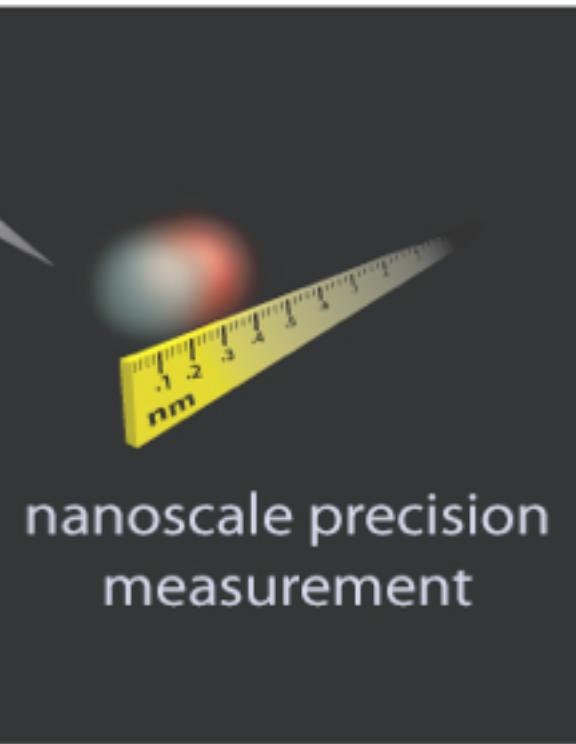
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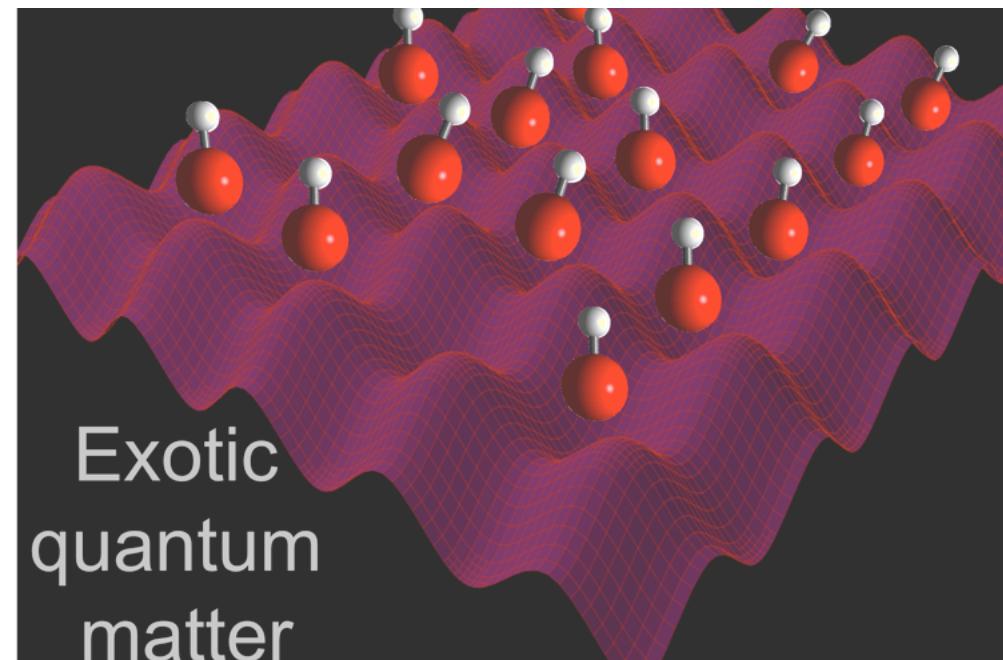
quantum computing

Possible Applications

ultracold quantum chemistry



nanoscale precision measurement



Exotic quantum matter

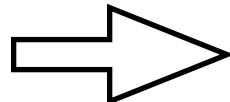
Outline

- * challenges
- * our approach - coherent transfer
- * making ro-vibronic ground-state polar molecules in a single quantum state
- * ultracold chemistry and dipolar collisions
- * outlook

Status of the quest for a Quantum Gas of Polar Molecules

$$PSD(n\Lambda^3) = 10^{-13} \text{ to } 10^{-15}$$

- * Direct cooling of molecules (Doyle, Meijer, Ye, Rempe, ...)

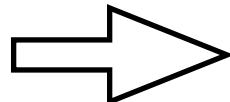


mK

make molecules

cooling

- * Photoassociation of ultracold atoms (DeMille, Stwalley, Bigelow, Weidmuller, ...)

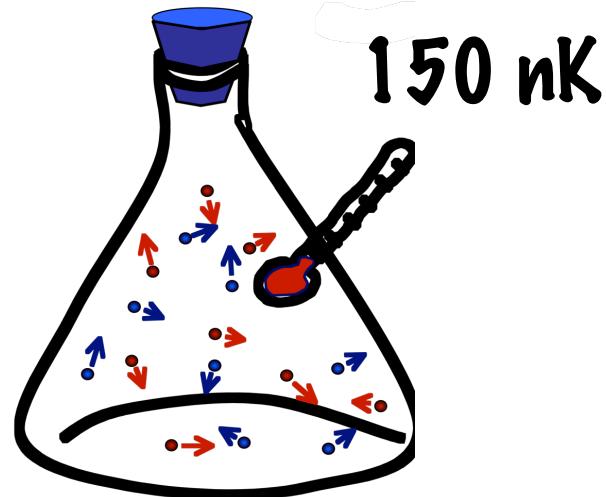


100 μK

cooling

make molecules

Starting from Ultracold Atoms

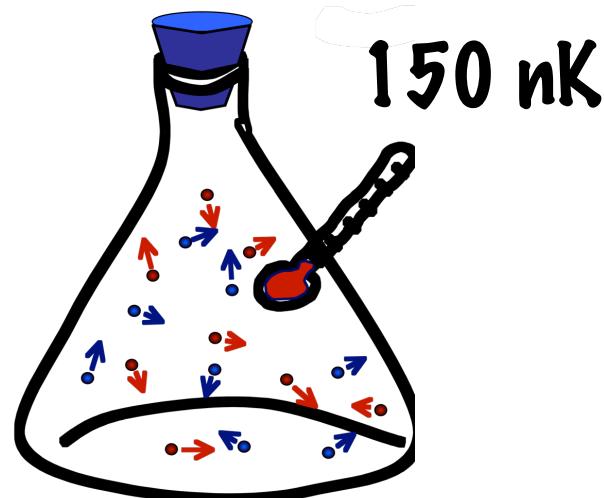


Two types of atoms

Rubidium
(boson)

Potassium
(fermion)

Starting from Ultracold Atoms

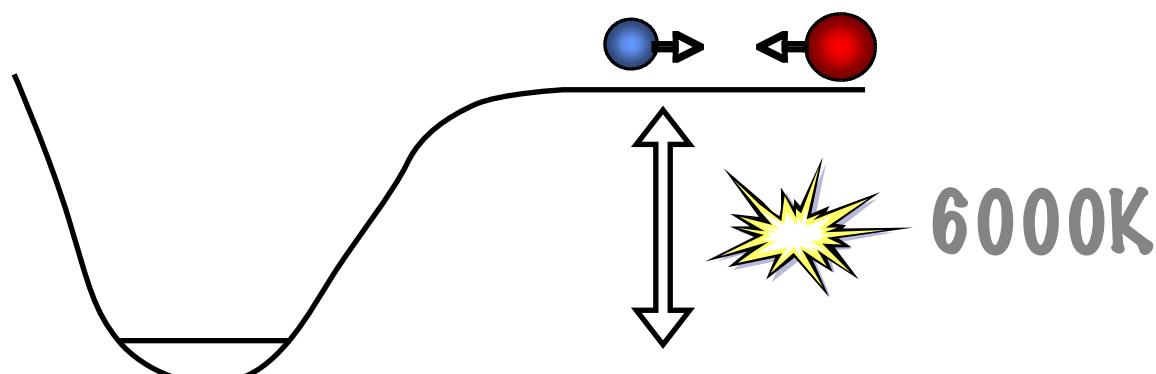


Two types of atoms

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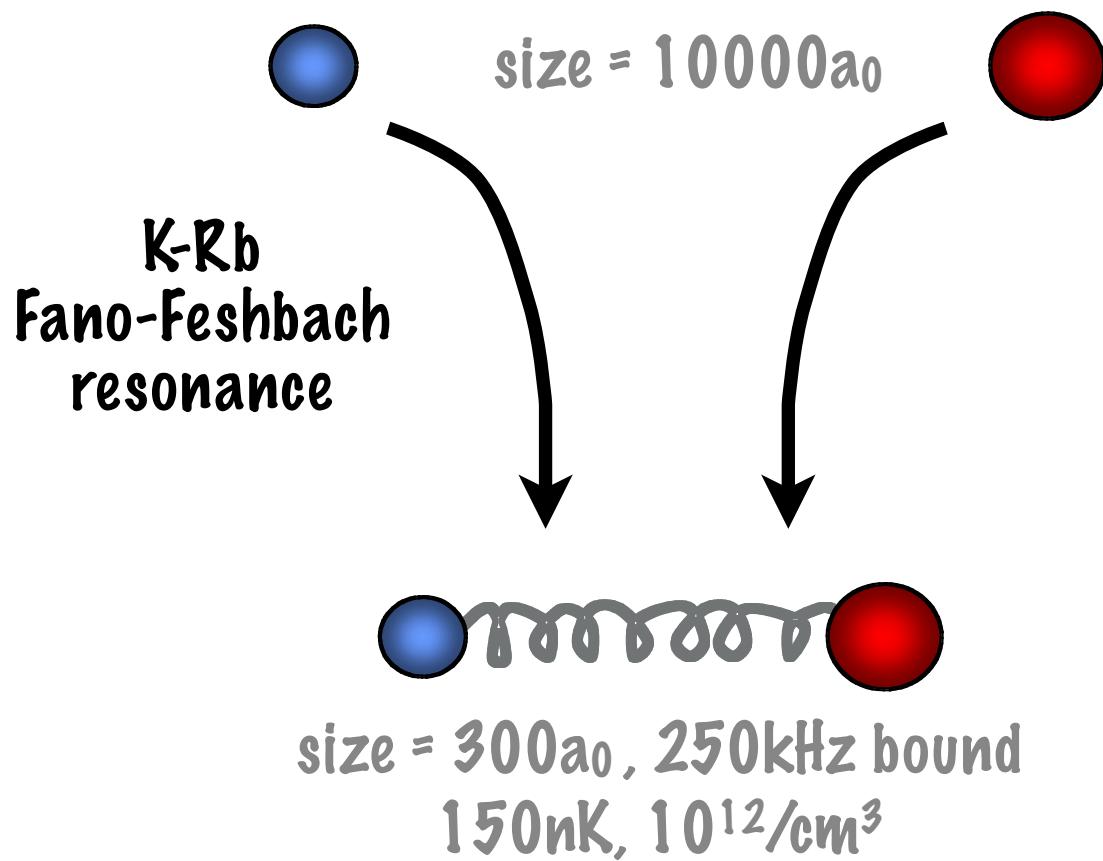
Potassium
(fermion)

challenges: efficient molecule creation while staying cold



Making KRb Feshbach Molecules

weakly bound “Feshbach” molecules can be created near a magnetic field tunable atomic scattering resonance



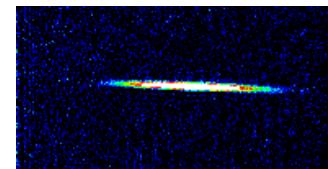
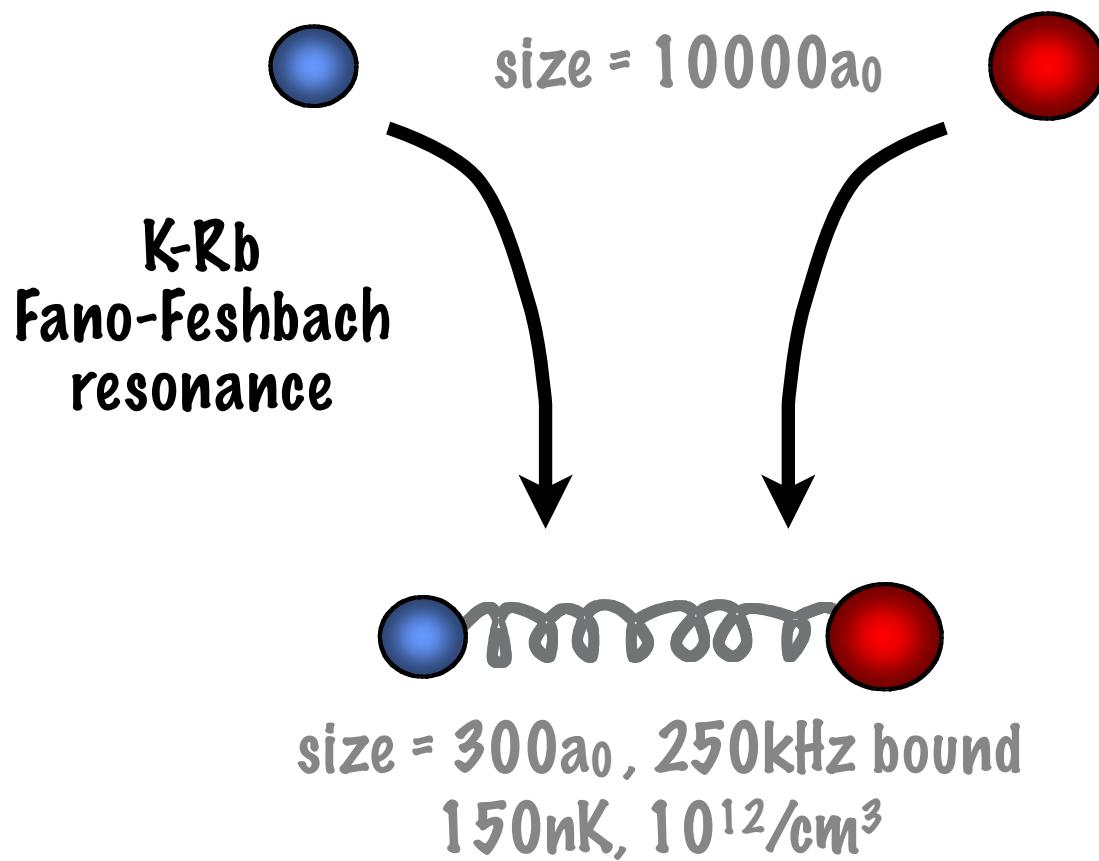
Related exp work:

- Inouye *et al.*, Nature **392**, 151 (1998)
Donley *et al.*, Nature **417**, 529 (2002)

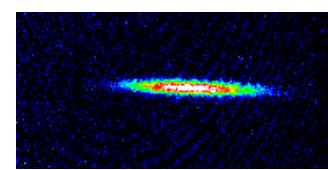
- Ospelkaus *et al.*, PRL **97**, 120402 (2006)
Zirbel, Ni *et al.*, PRL **100**, 143201 (2008)
Zirbel, Ni *et al.*, PRA **78**, 013416 (2008)

Making KRb Feshbach Molecules

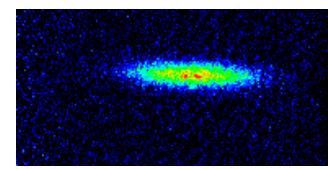
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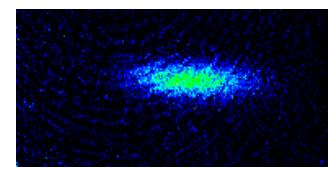
TOF = 1 ms



TOF = 3 ms



TOF = 6 ms



TOF = 9 ms

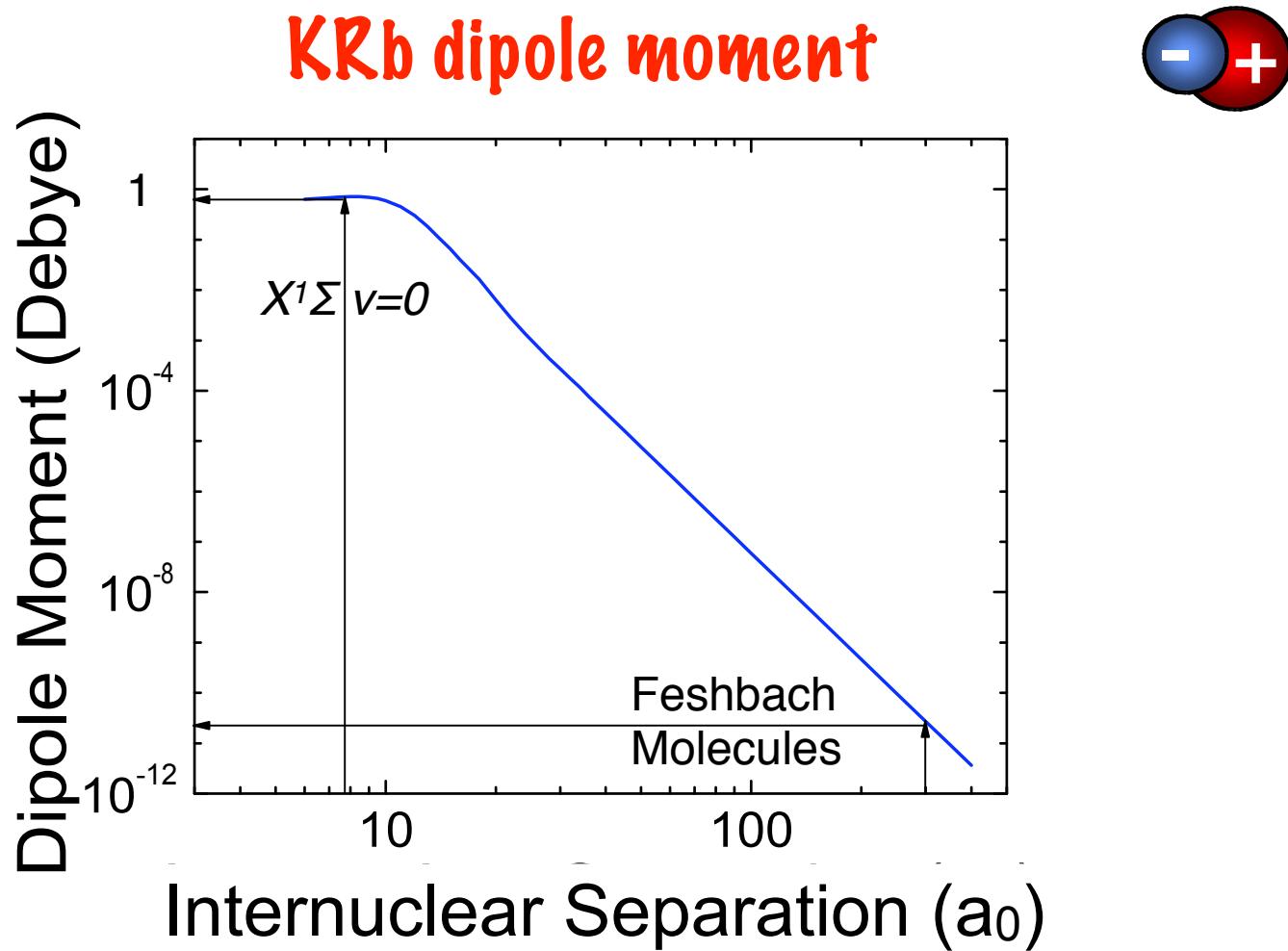
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Shrink the Molecules

small, tightly bound, polar molecules

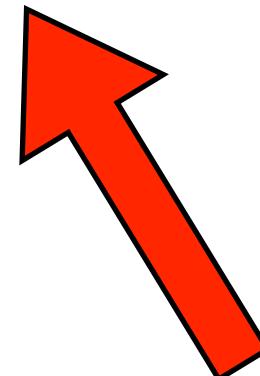
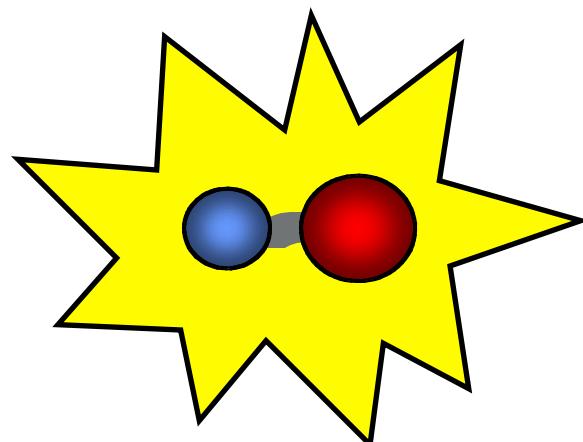


Kotochigova *et al.*, Phys. Rev. A **68**, 022501 (2003)

making molecules Ultracold and Polar

In A Single Internal Quantum State

light carries away
binding energies



Laser light



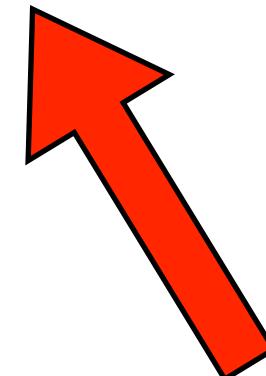
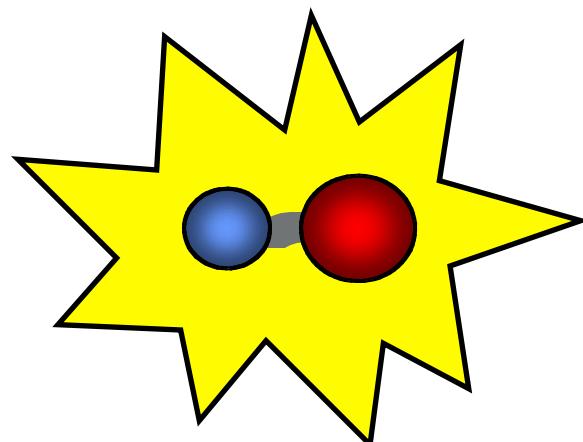
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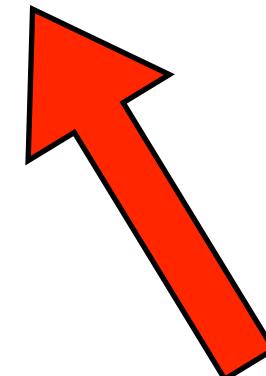
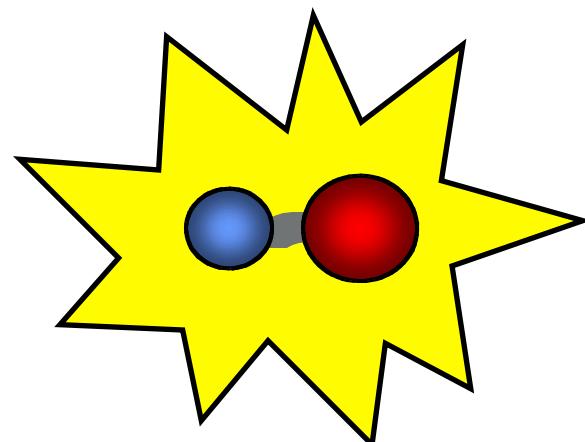
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making molecules Ultracold and Polar

In A Single Internal Quantum State

light carries away
binding energies

non-Coherent Process
recoil heating (160 nK)
multiple final states



Related work:

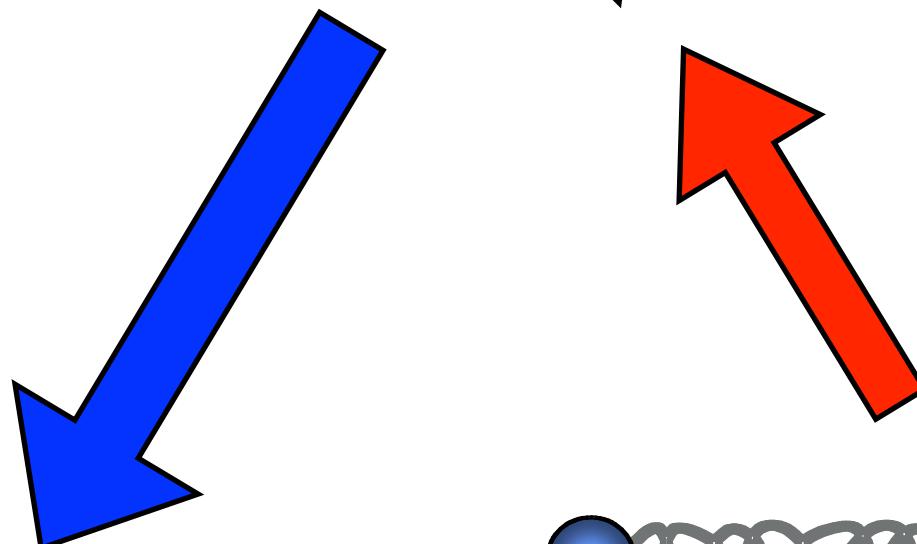
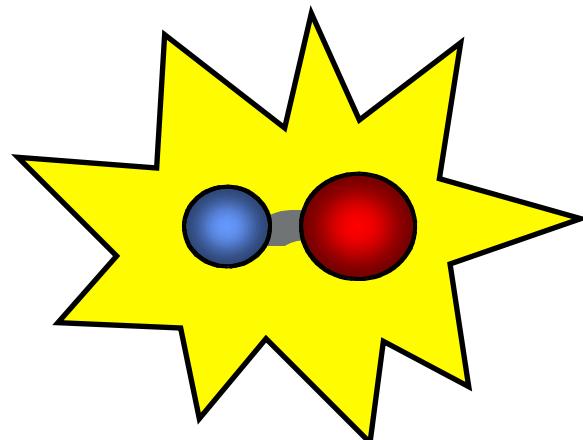
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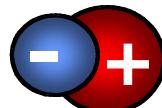
making molecules Ultracold and Polar

In A Single Internal Quantum State

light carries away
binding energies



Laser light



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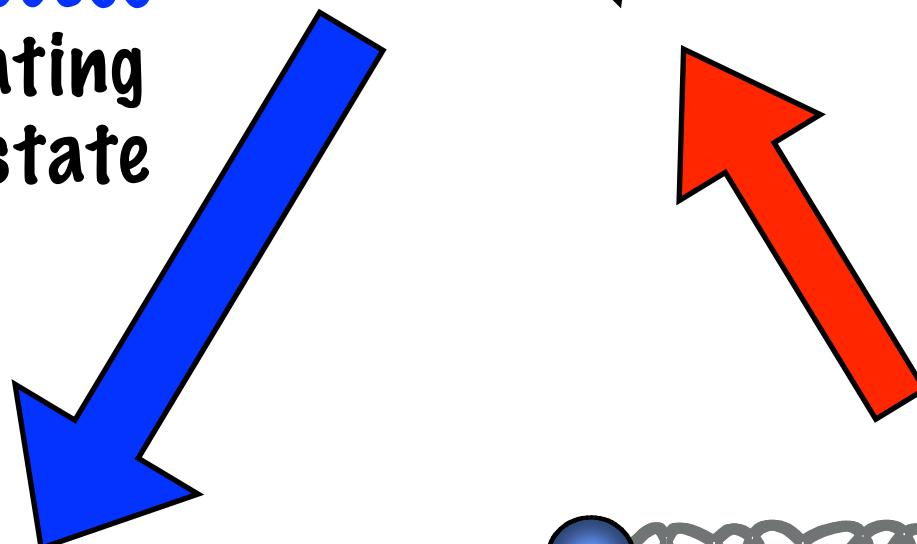
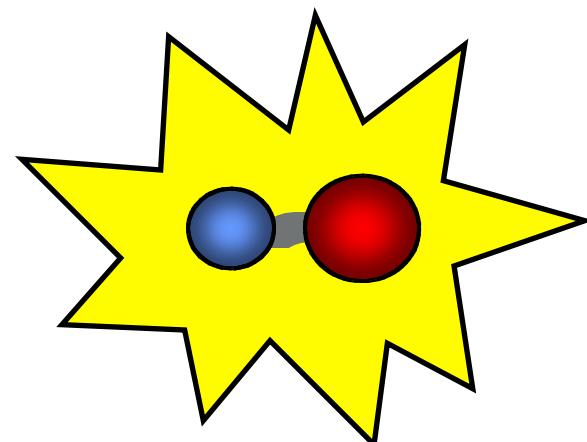
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making molecules Ultracold and Polar

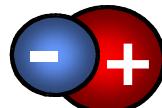
In A Single Internal Quantum State

light carries away
binding energies

Coherent Process
no recoil heating
single final state



Laser light



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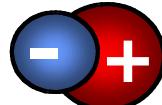
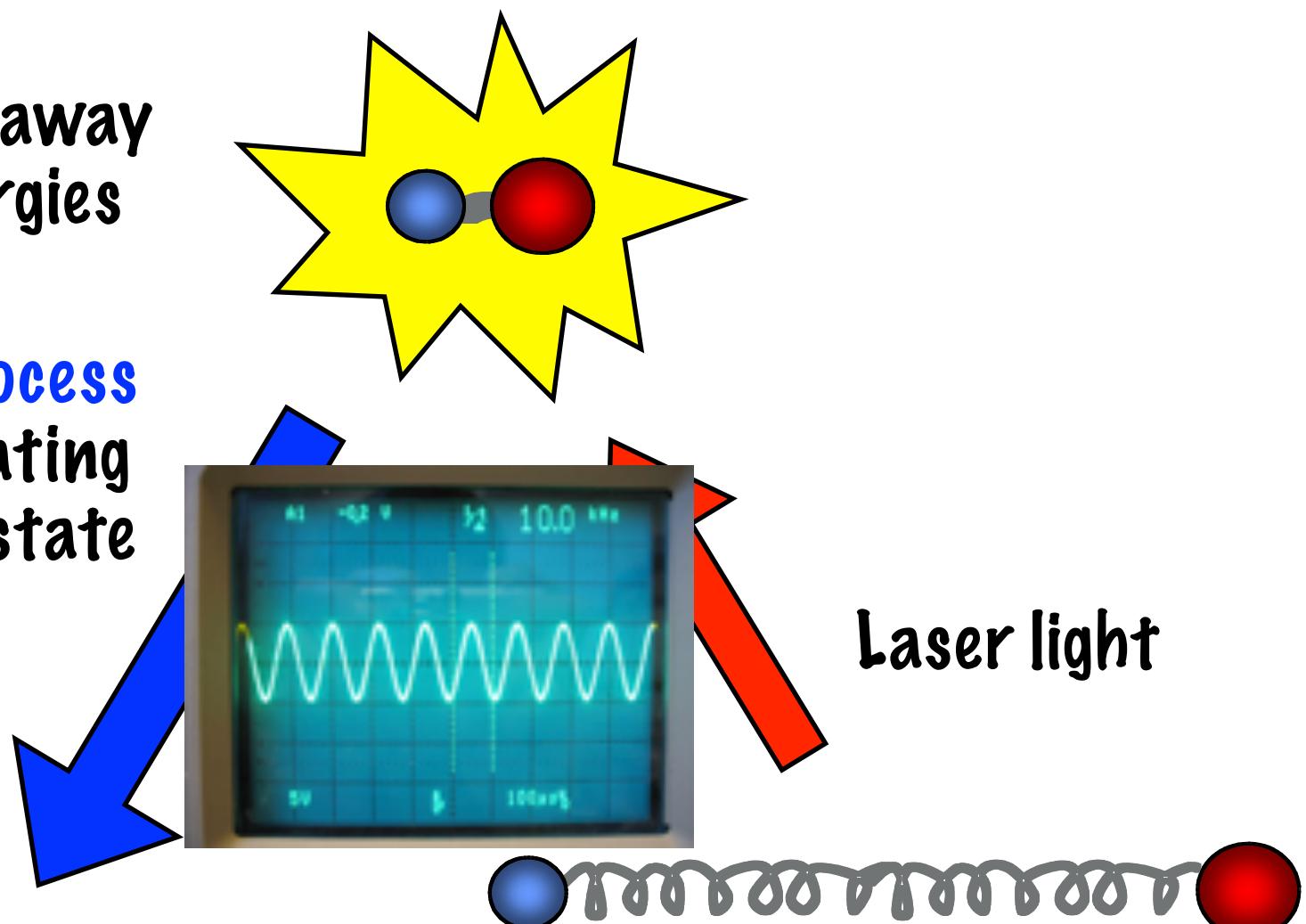
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making molecules Ultracold and Polar

In A Single Internal Quantum State

light carries away
binding energies

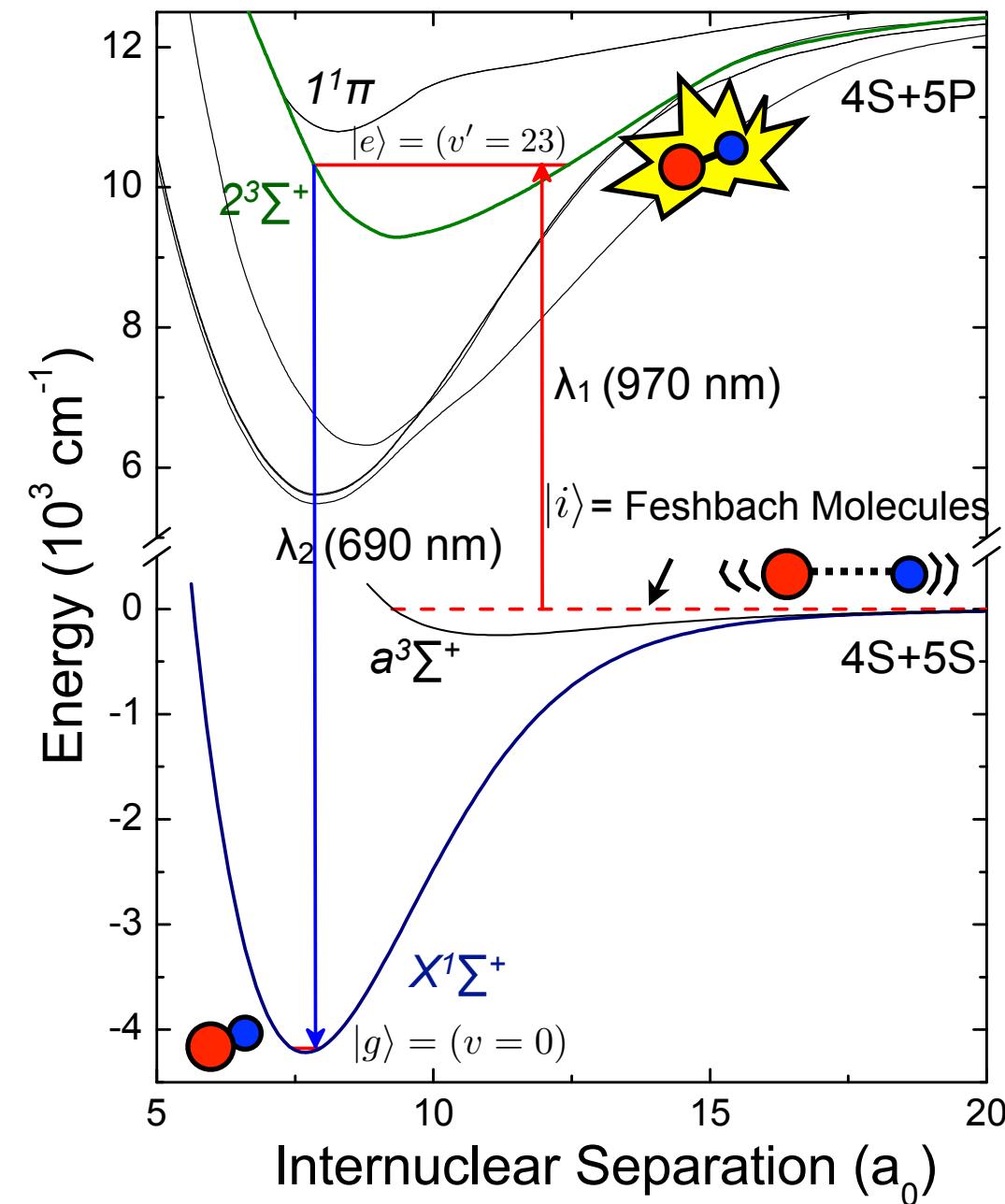
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One Magic Bridge

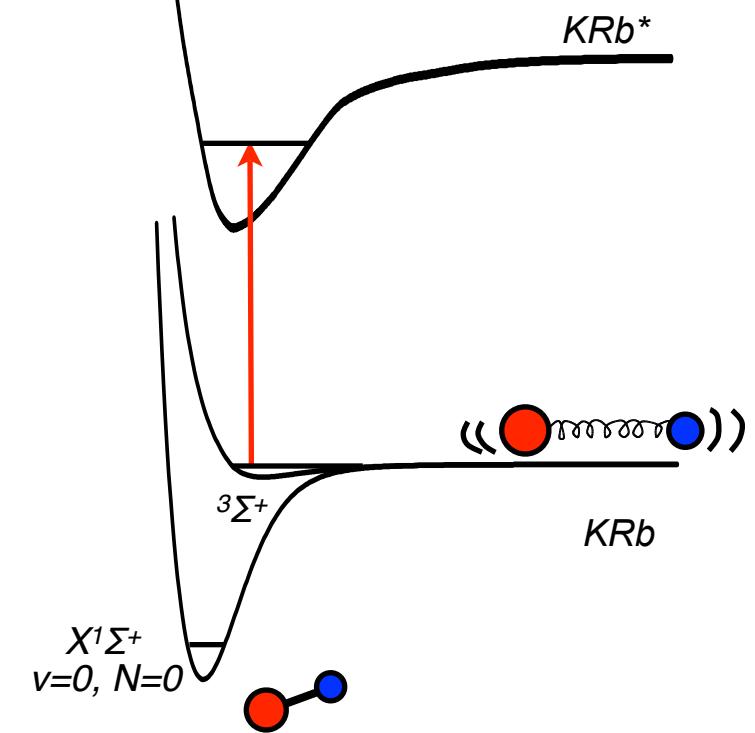


Large Franck-Condon factors
non-trivial triplet-singlet mixing

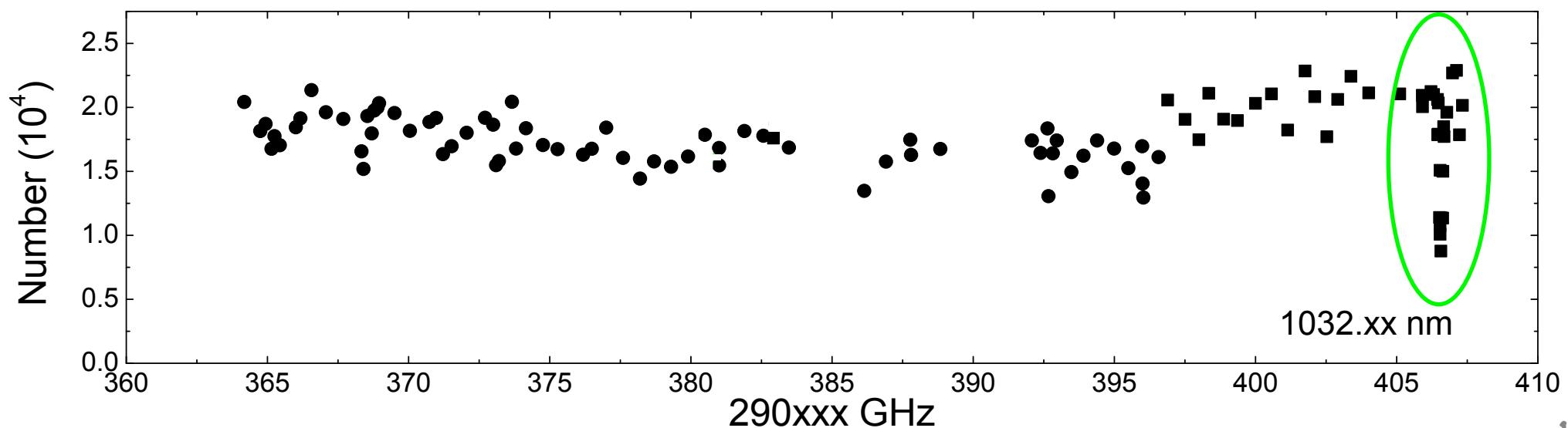
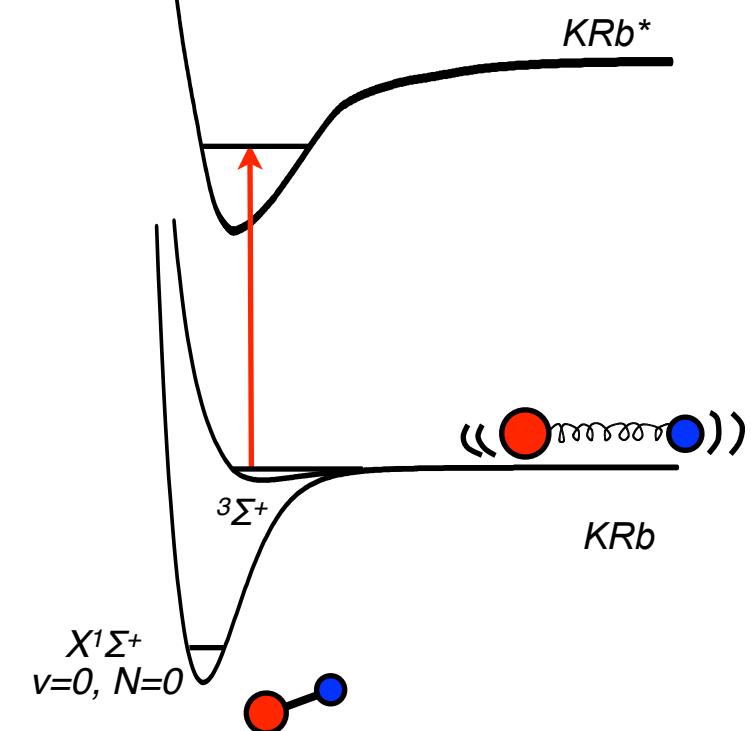
Ni *et al.*, Science **322**, 231 (2008)

Related photoassociation work:
Sage *et al.*, PRL **94**, 203001 (2005)

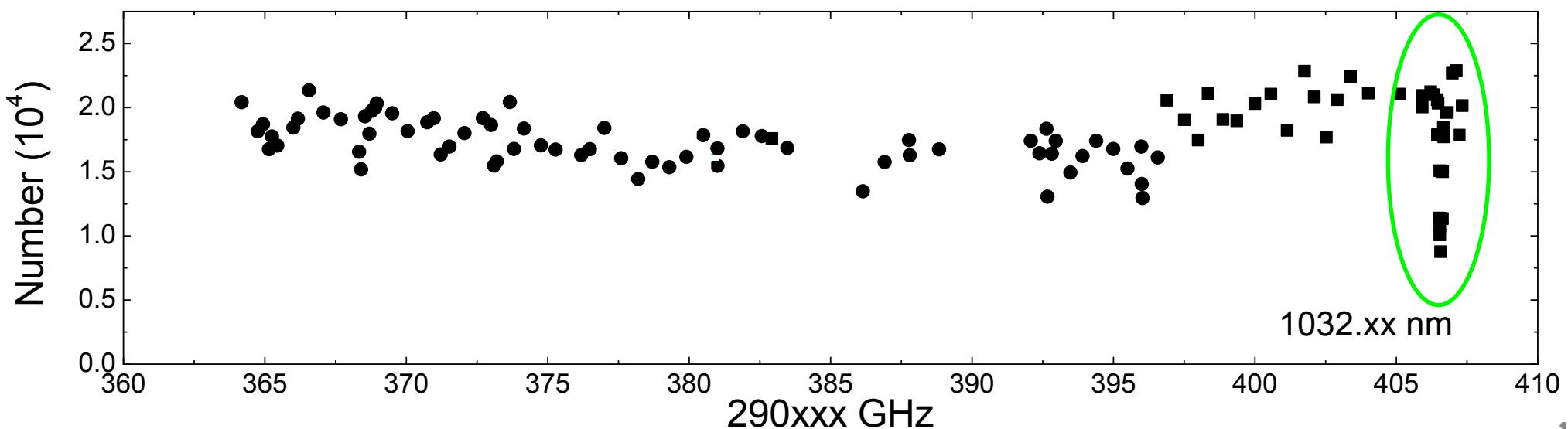
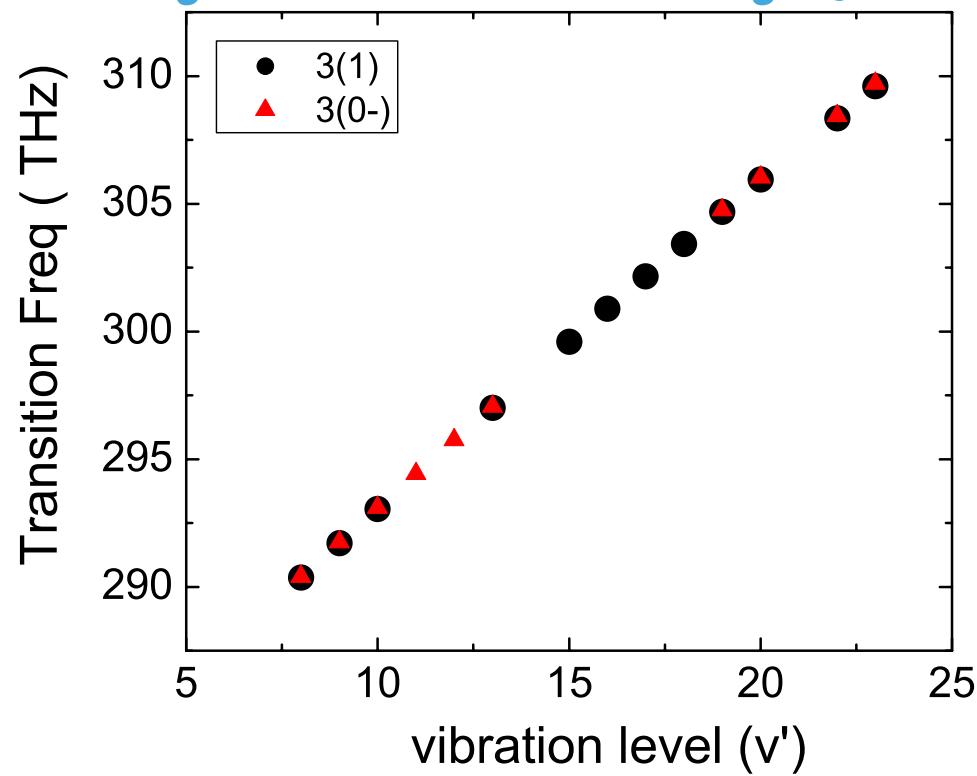
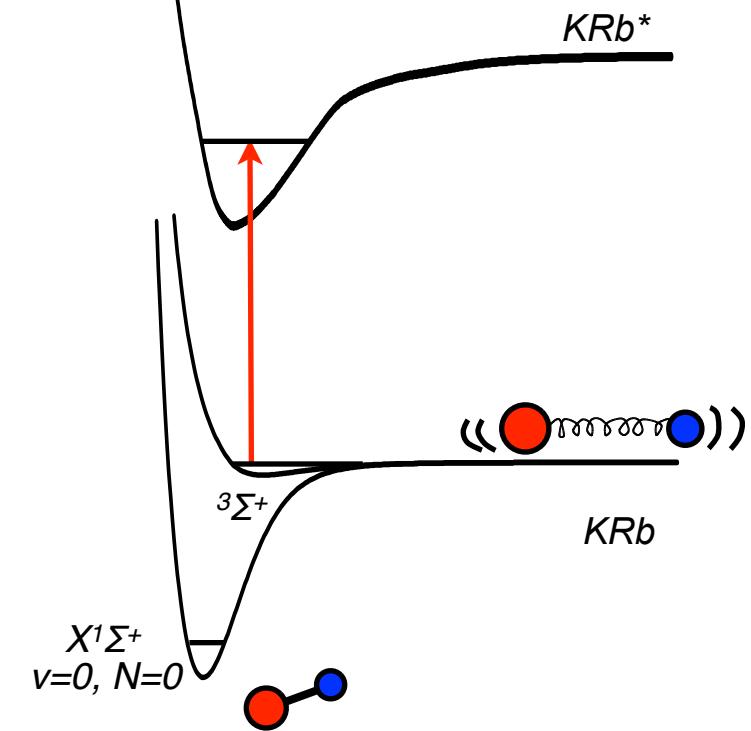
One-Photon Spectroscopy



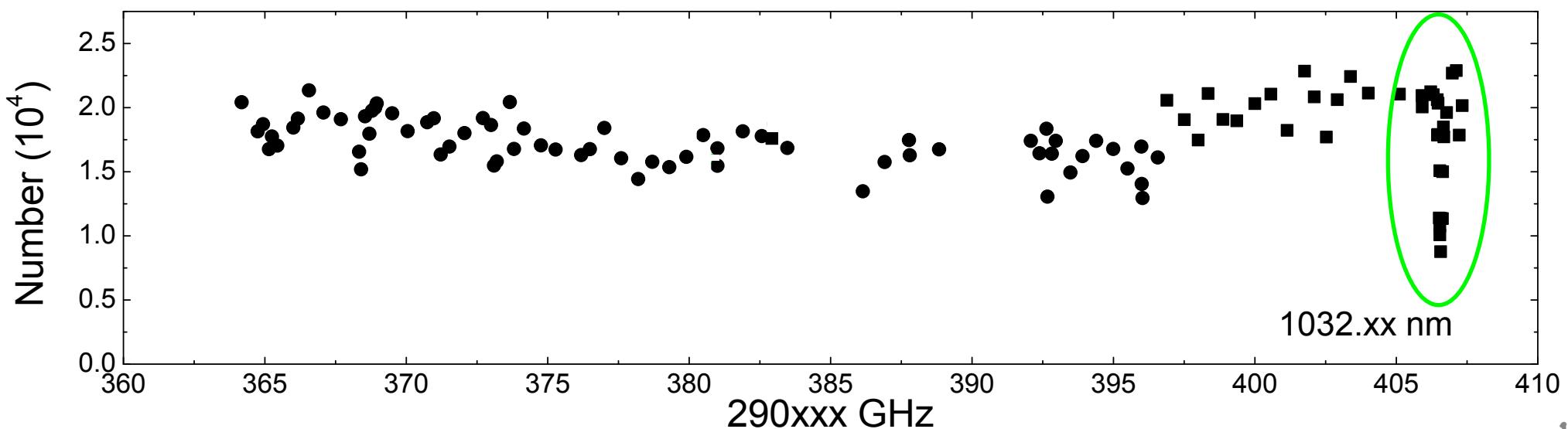
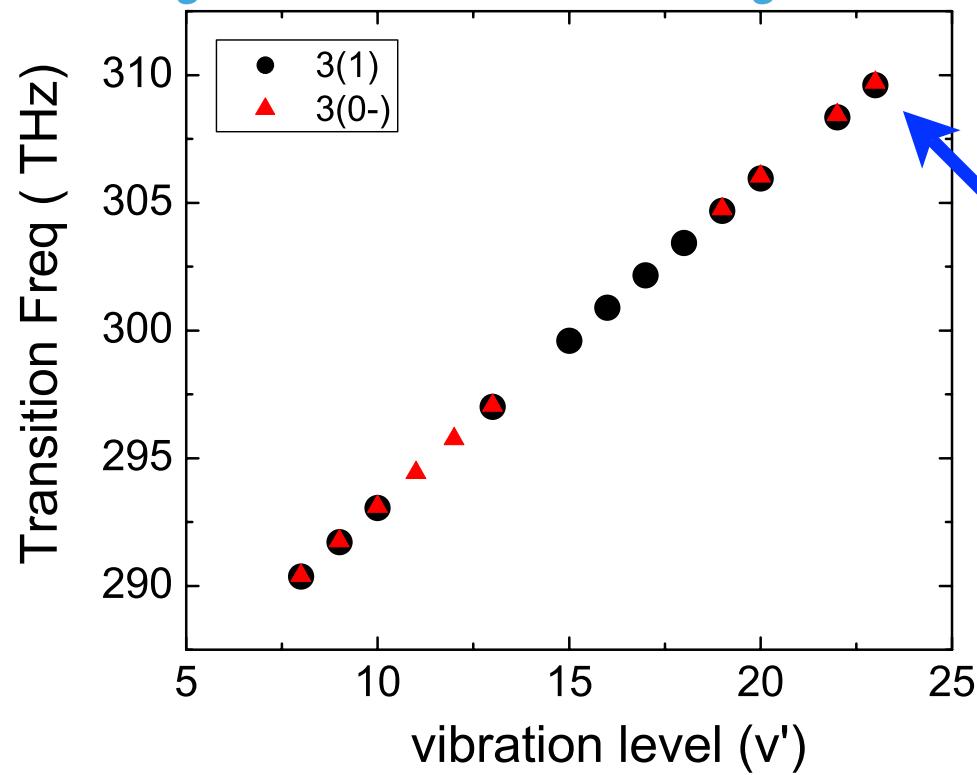
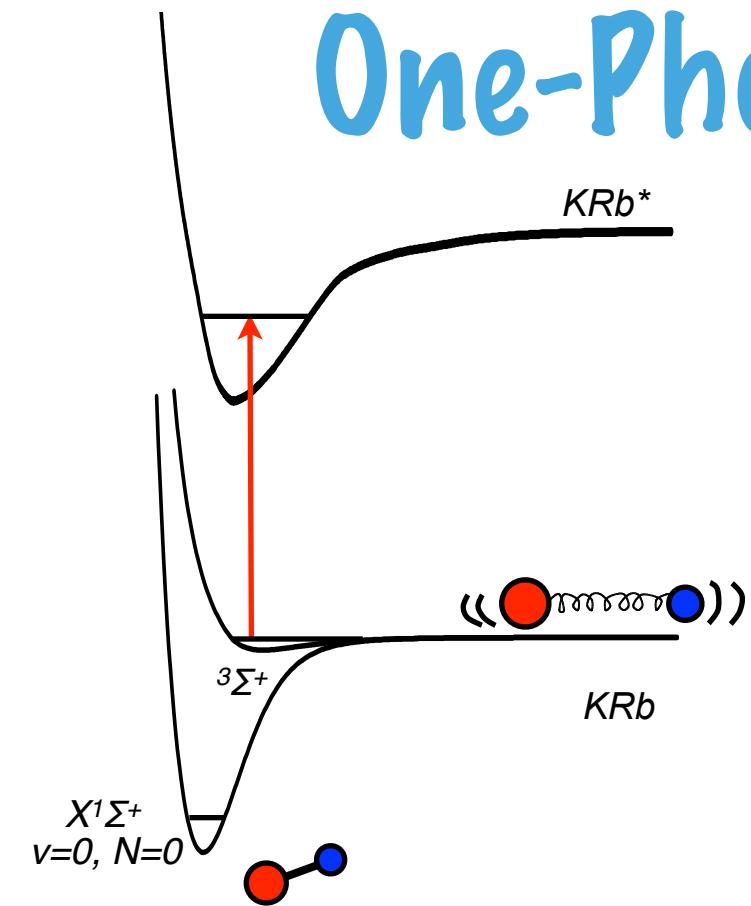
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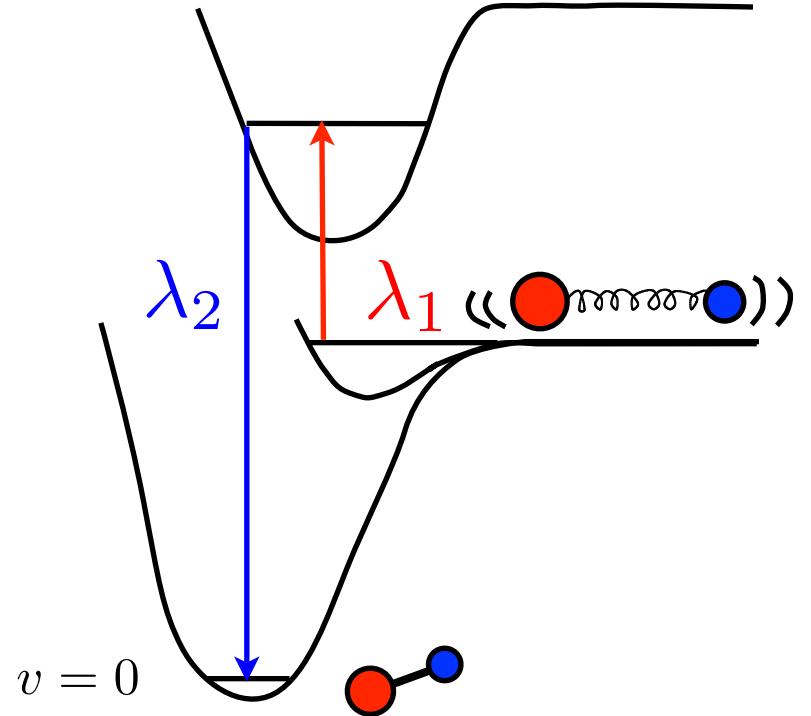
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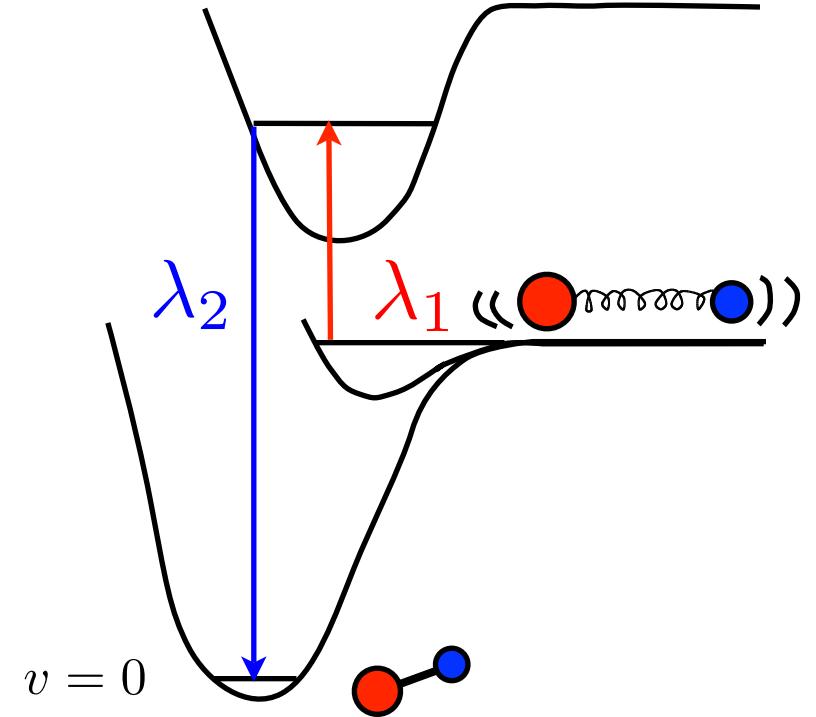
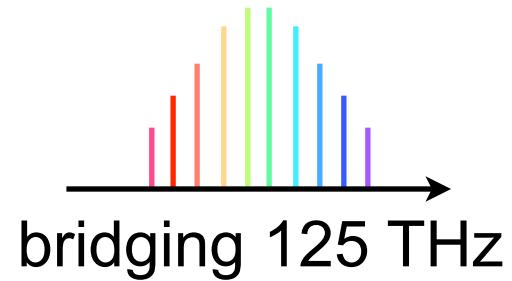
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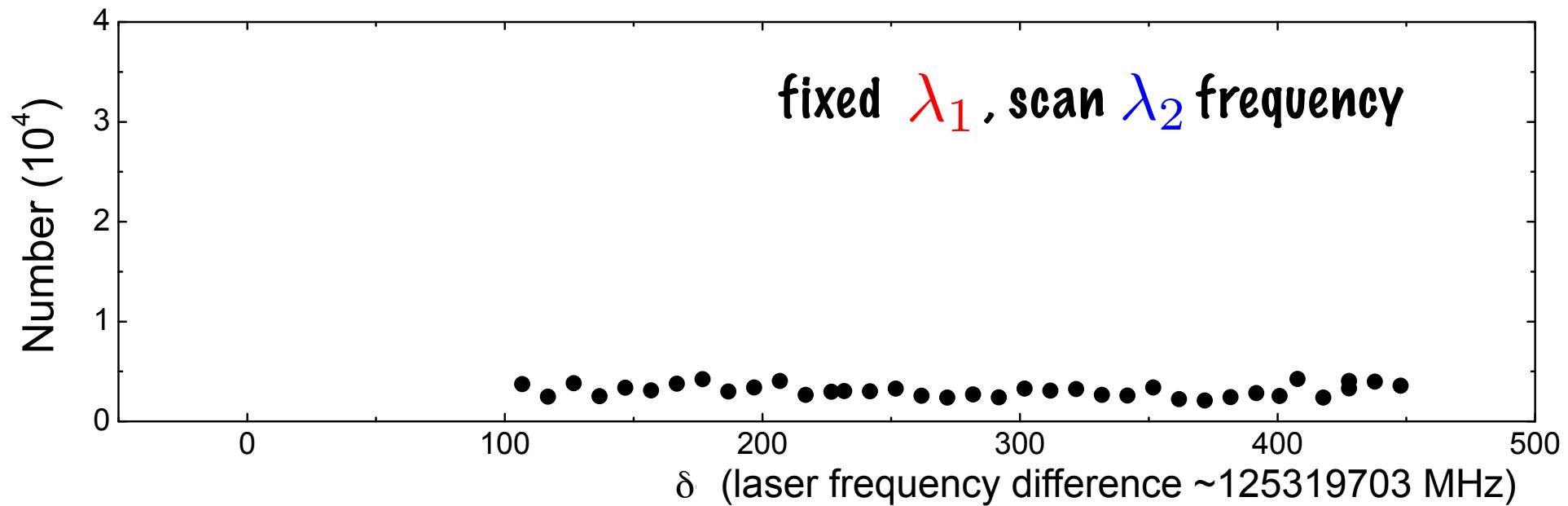
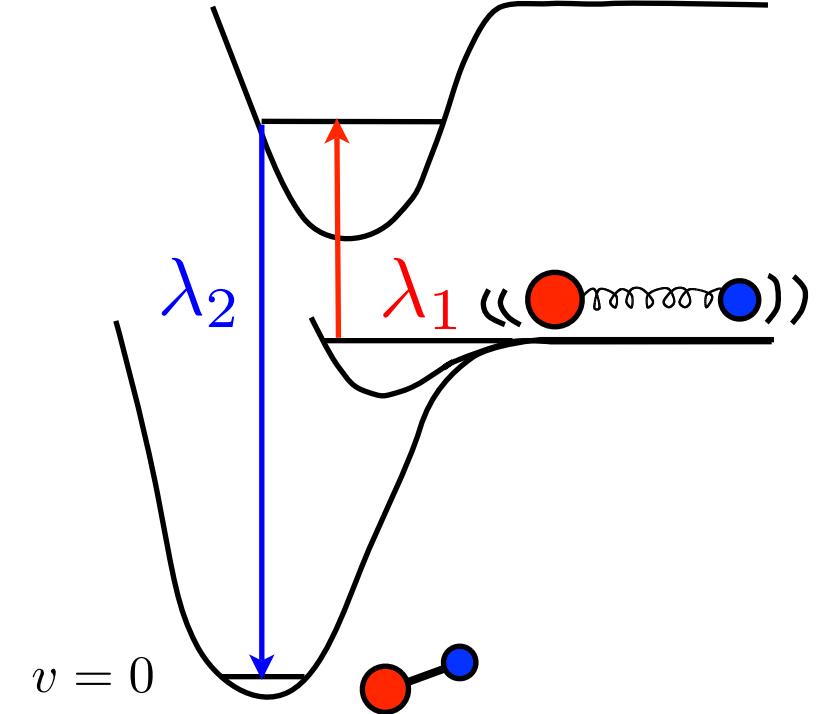
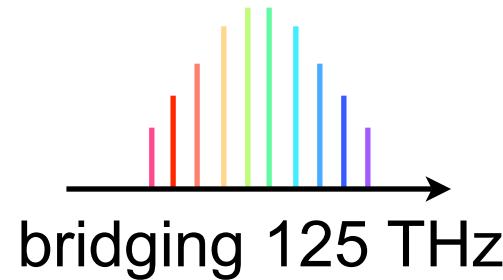
Dark Resonance



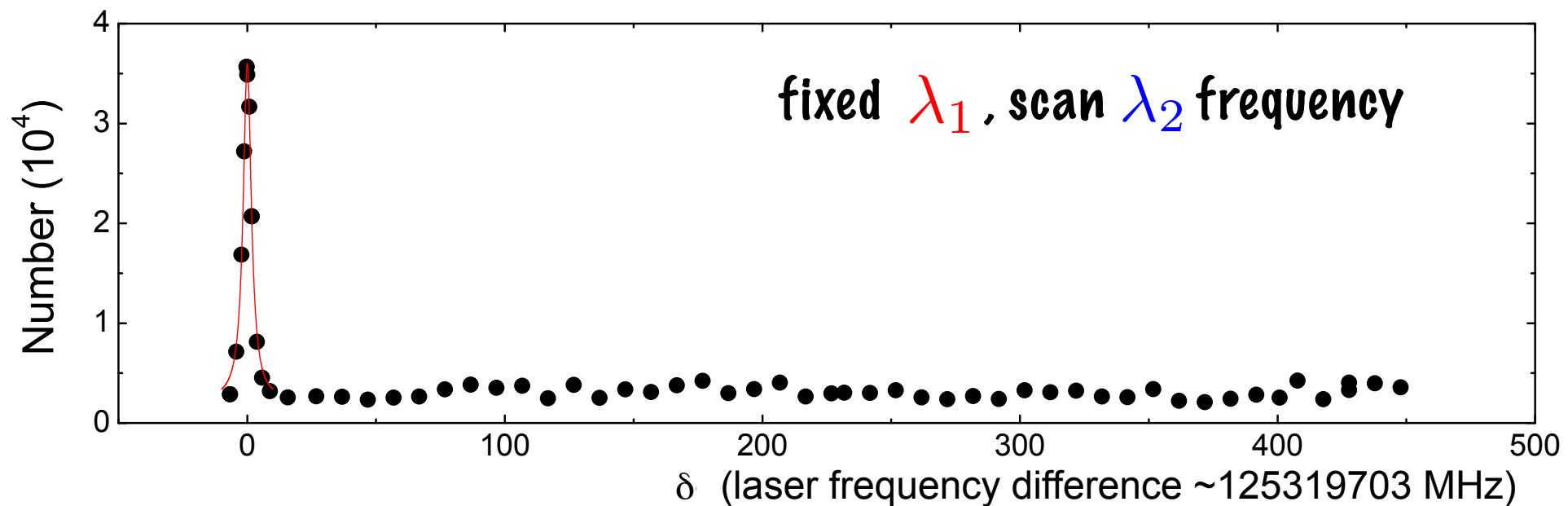
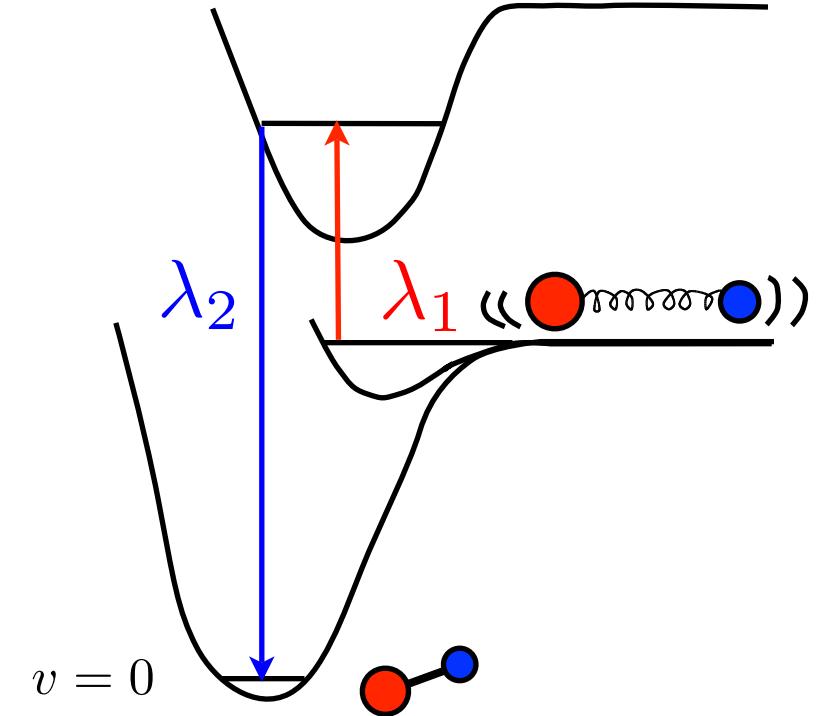
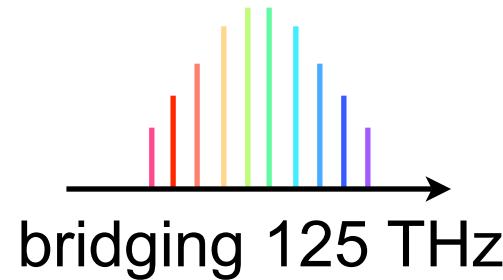
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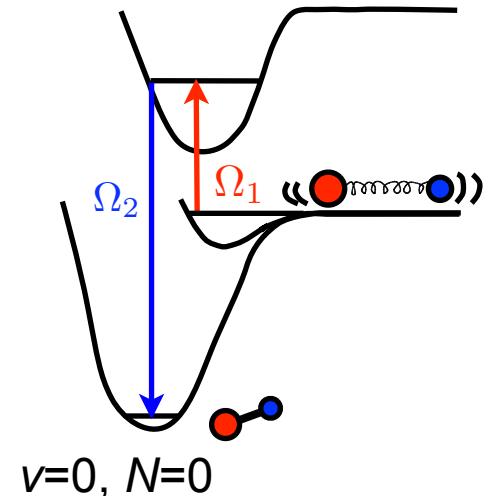
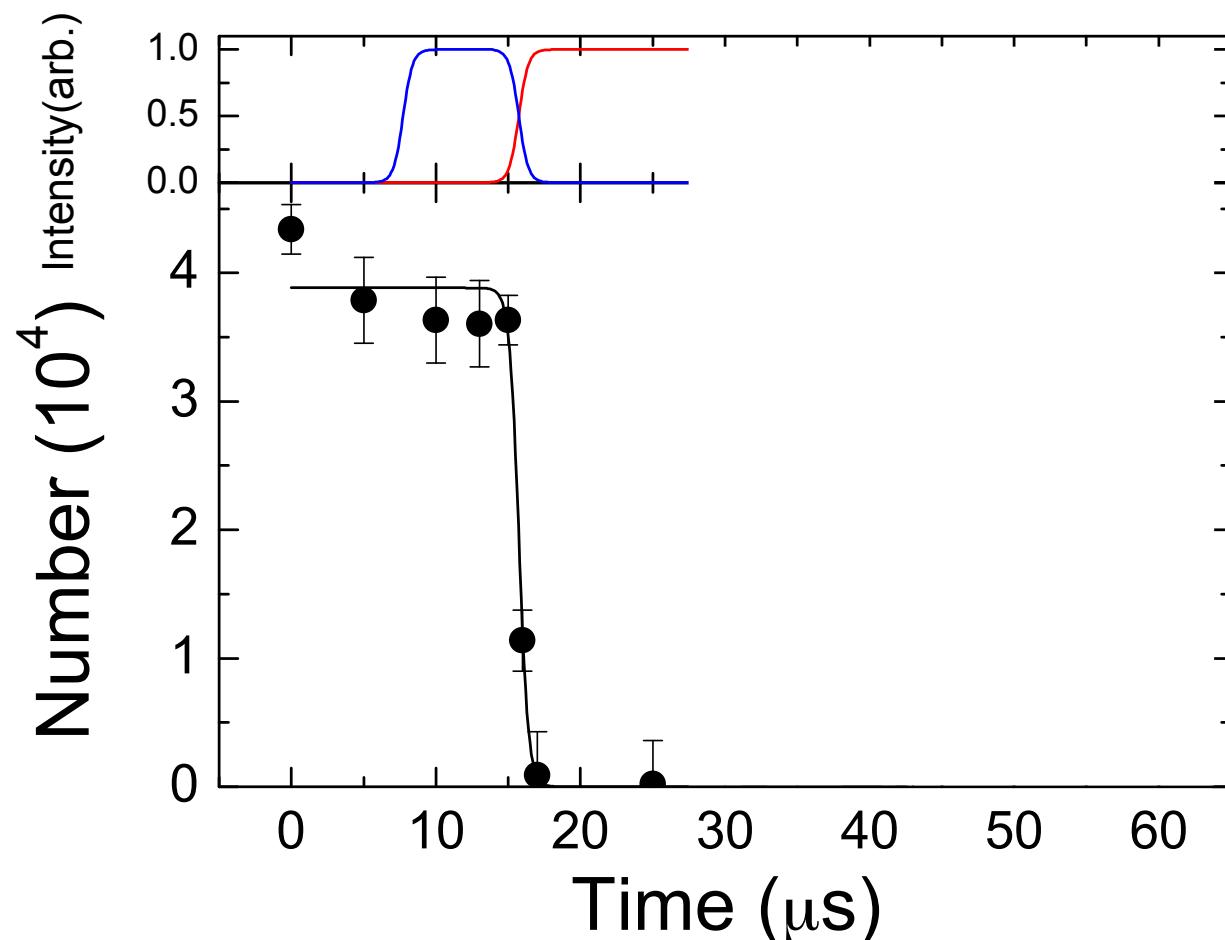


Dark Resonance



STIRAP

STImulated Raman Adiabatic Passage
4 μ s one-way transfer



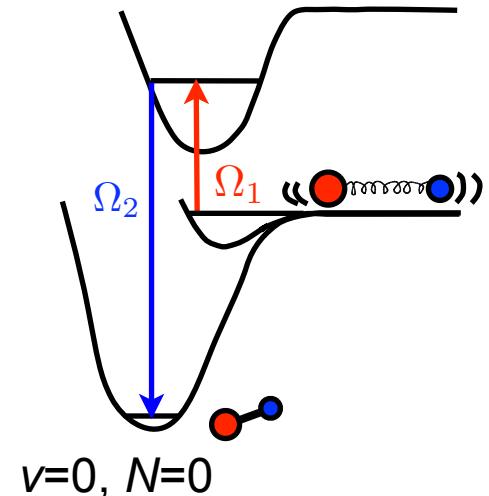
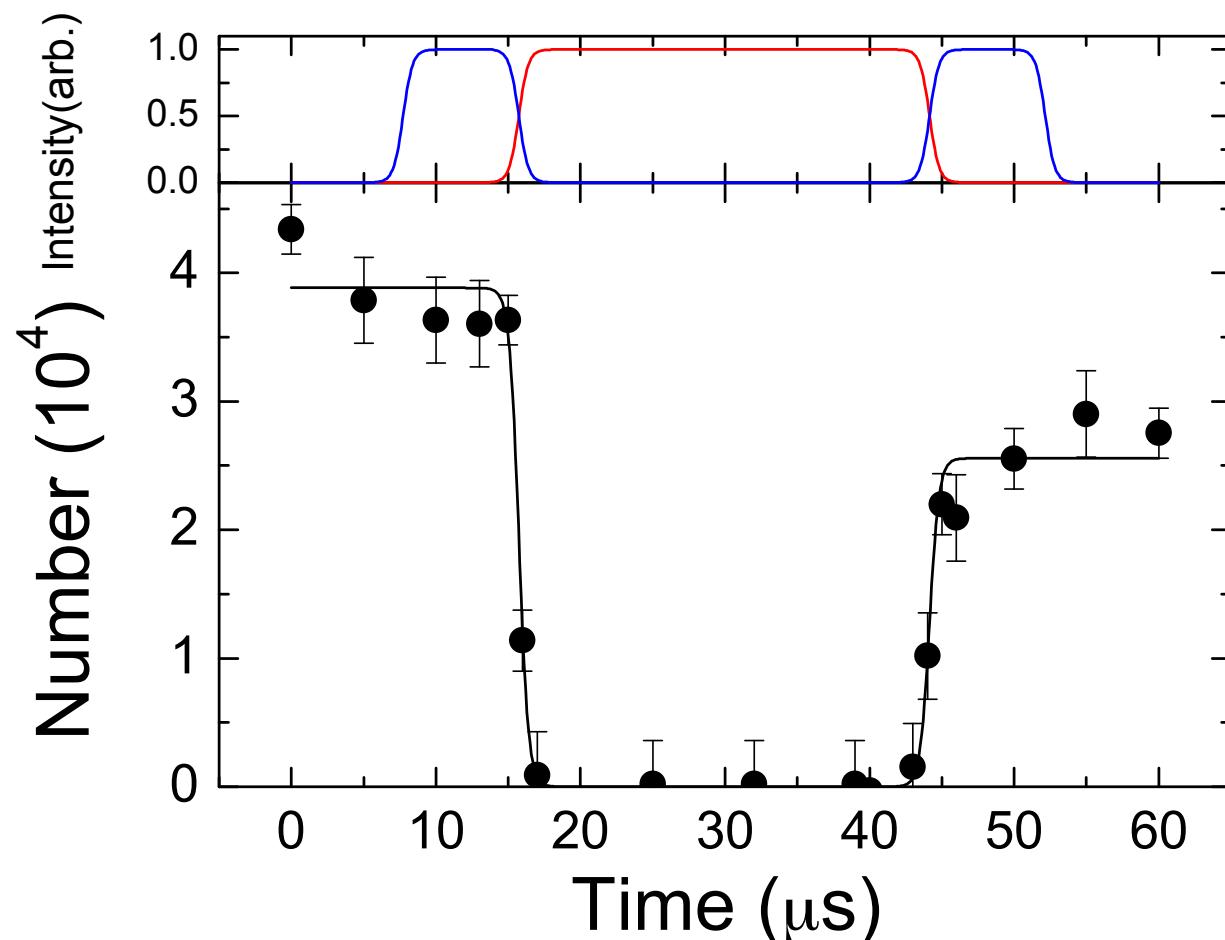
>80-90%
one-way
efficiency

STIRAP review: K. Bergmann et al., RMP **70**, 1003-1025 (1998)

Related work for non-polar molecules, Cs₂ and Rb₂ (Innsbruck, 2008)

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Trapped fermionic molecules

$\text{PSD} = 0.06$

peak density = $10^{12}/\text{cm}^3$

temperature = 160 nK

molecules in a single and lowest
rotational, vibrational, and electronic
ground state!

11 orders of magnitude improvement from previous results

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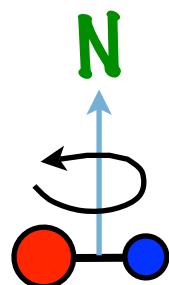
what about hyperfine structure?

Reaching the Lowest Hyperfine State

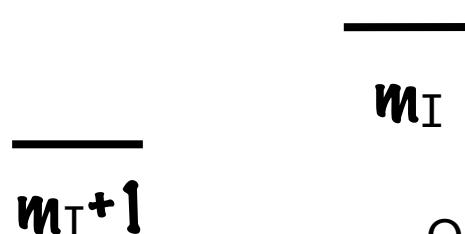
$v=0, N=0$, electronic orbital=0, electronic spin =0
but with nuclear spins:

36 total states

N=1



N=0



Reaching the Lowest Hyperfine State

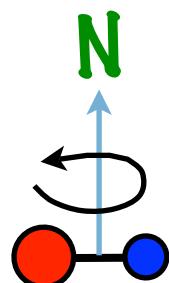
$v=0, N=0$, electronic orbital=0, electronic spin =0
but with nuclear spins:

36 total states

>90%

<10% admixture

$$N=1 \quad m_F = m_N - 1 + m_I + 1 \longrightarrow m_N + m_I$$



strong nuclear electric quadrupole interactions

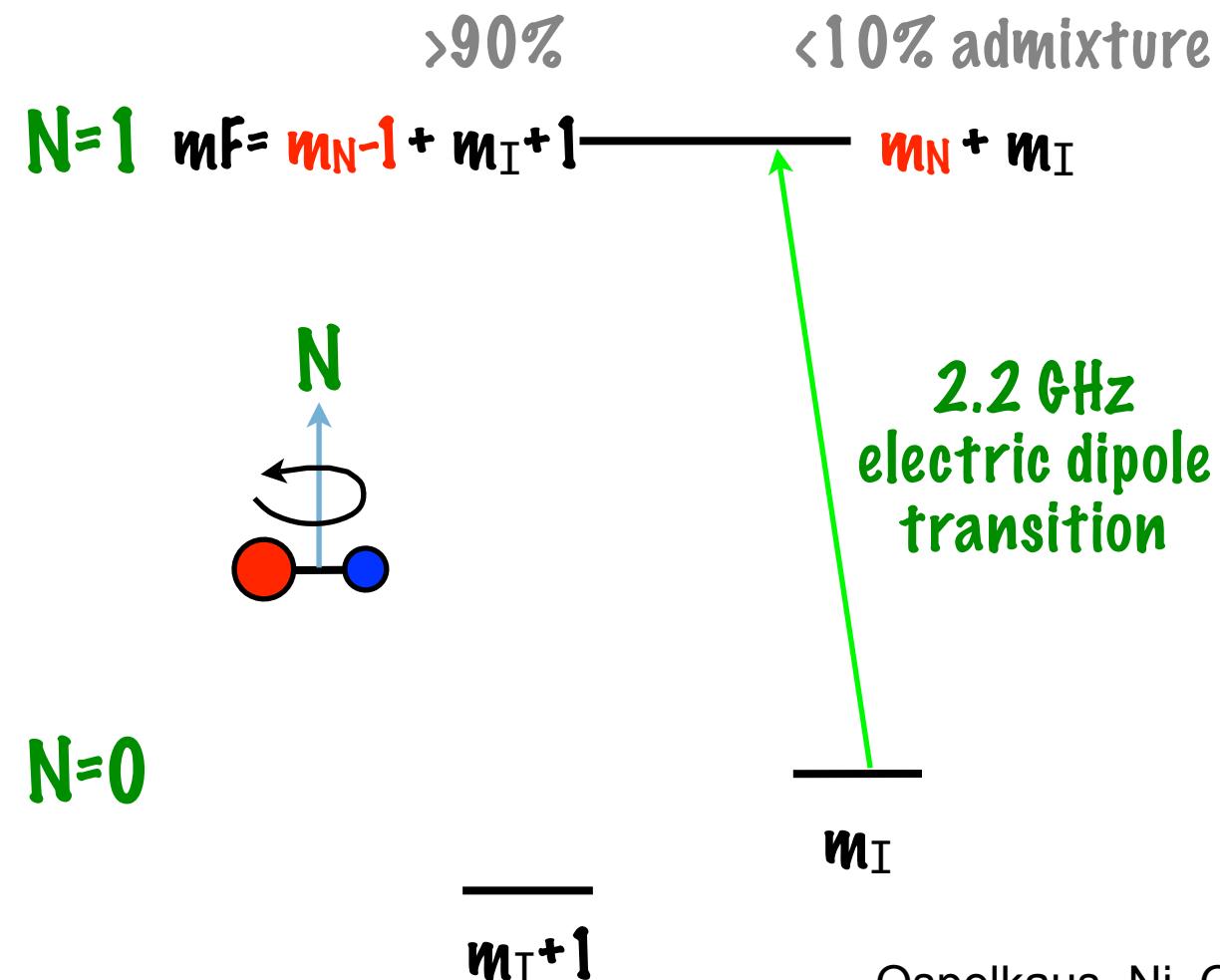
$N=0$

$$\frac{m_I}{m_{I+1}}$$

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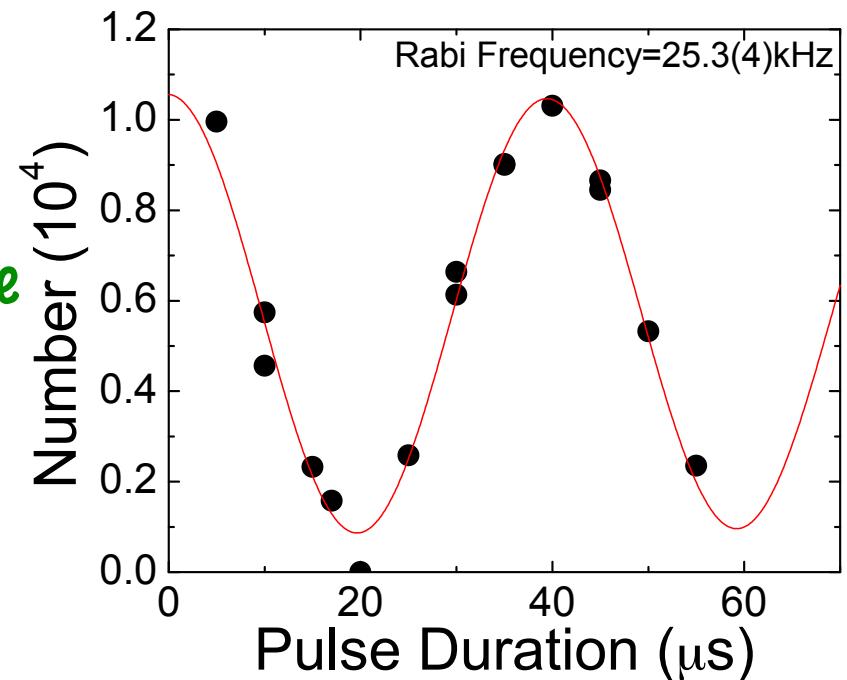
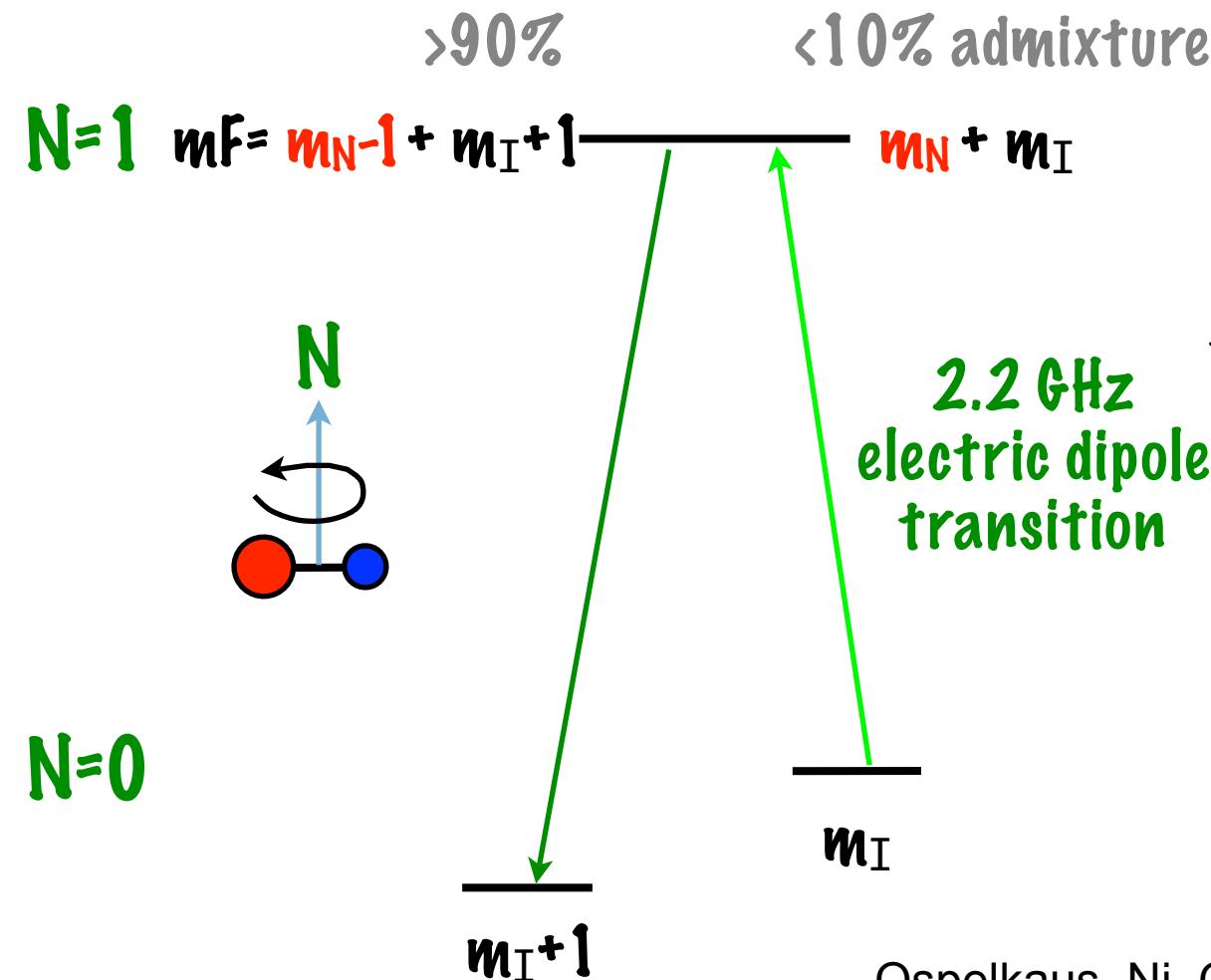


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$v=0, N=0$, electronic orbital=0, electronic spin =0
but with nuclear spins:

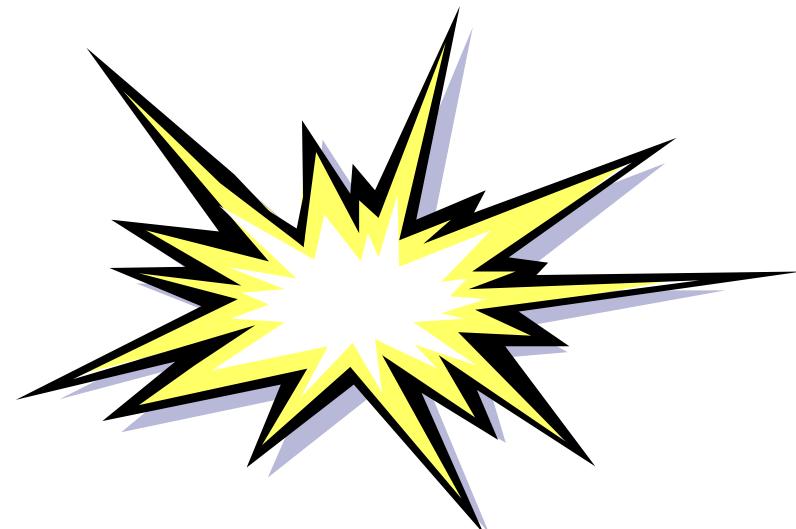
36 total states



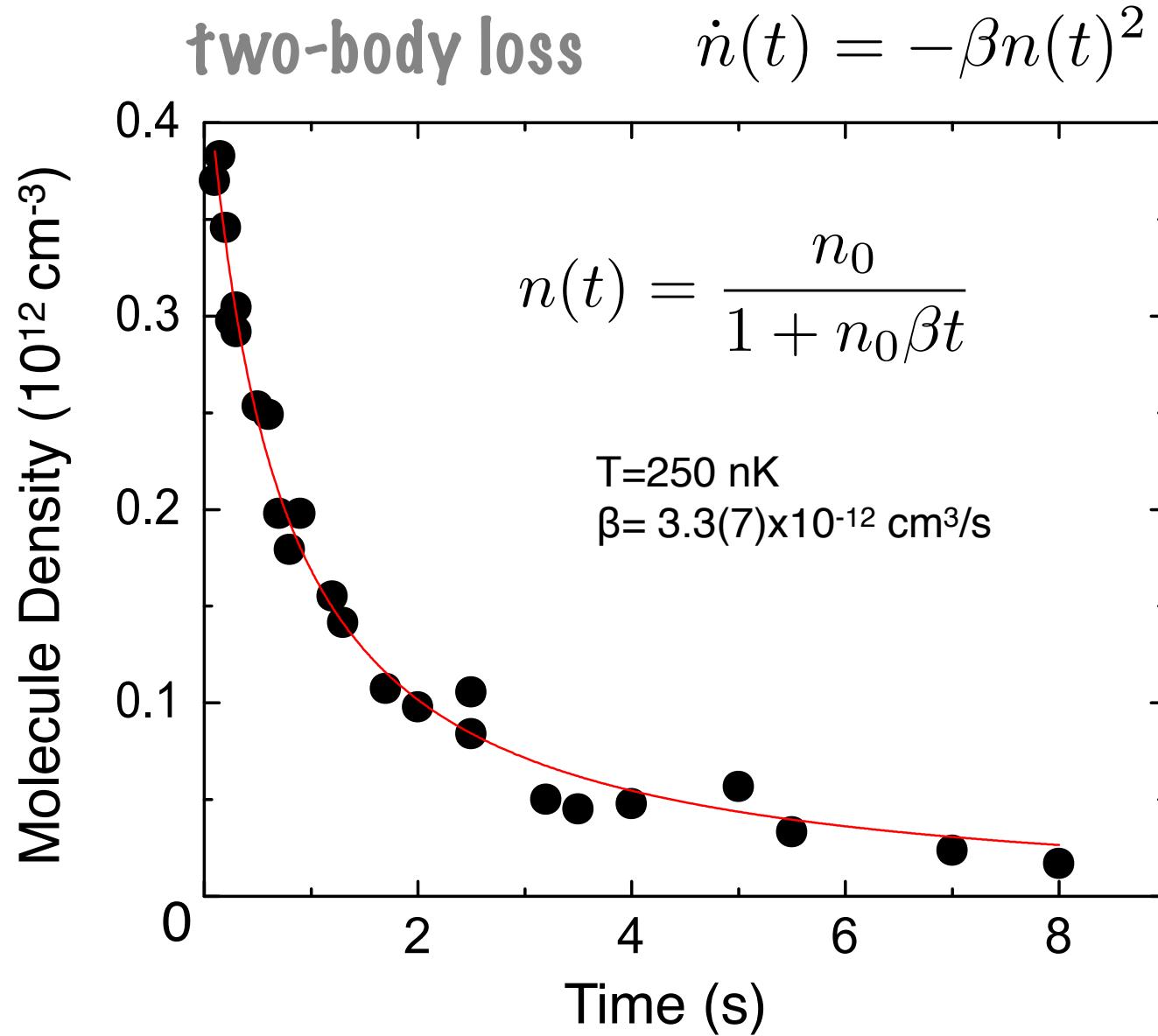
molecules in the lowest energy state (electronic, vibrational, rotational, and hyperfine)

can they decay?

what determines the rate?

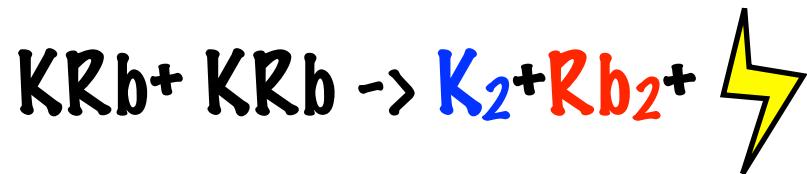


Long-Lived Molecules...



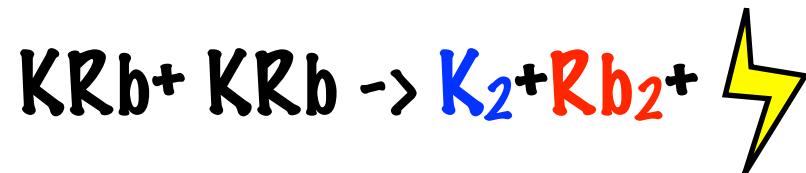
Ultracold Chemistry?!

possible exothermic reaction



Ultracold Chemistry?!

possible exothermic reaction

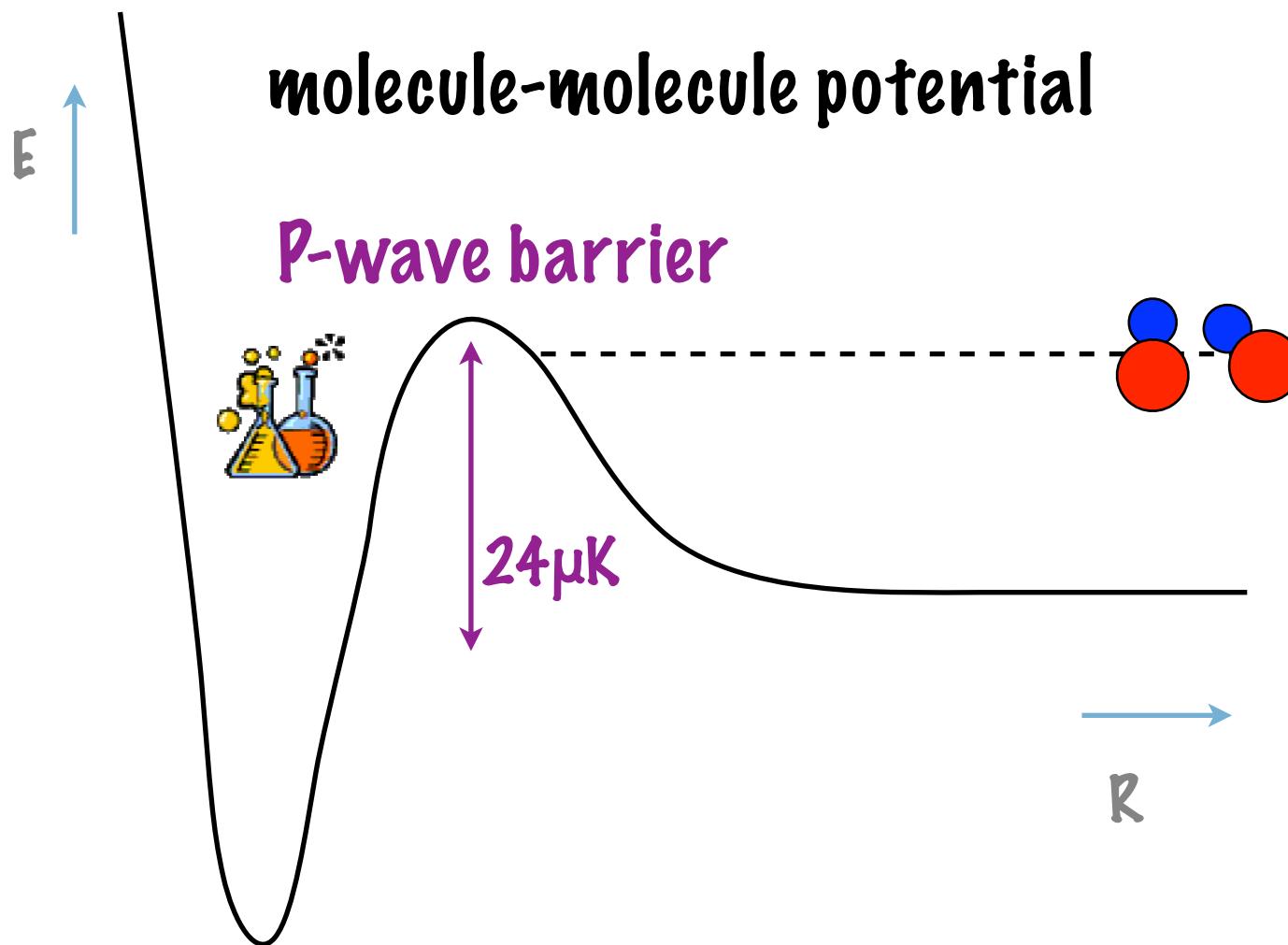


but our molecules are spin-polarized fermions...
collisions should be suppressed at ultralow T

Ultracold Fermi Gas of Molecules

Wigner Threshold Law
for inelastic collisions

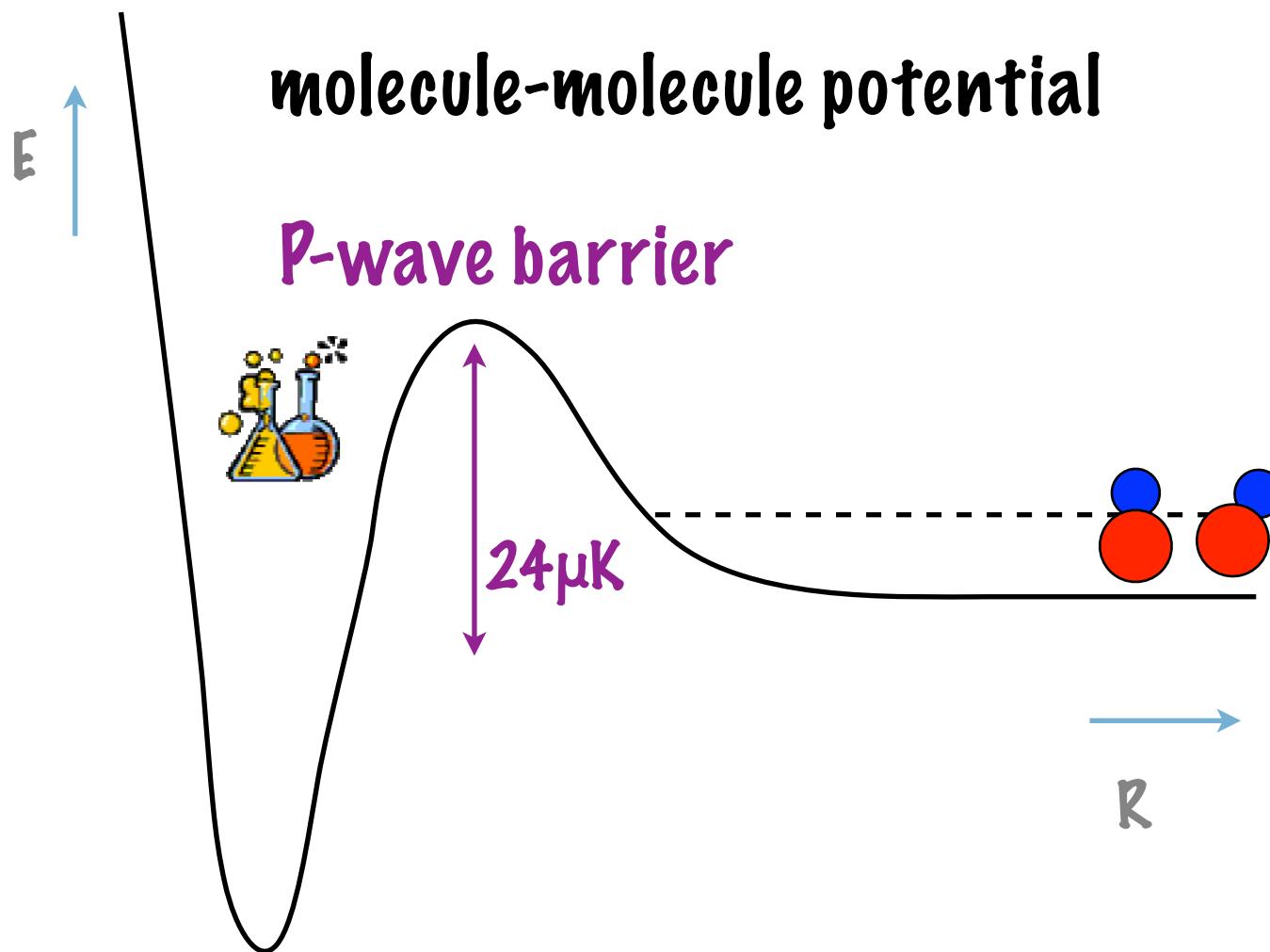
$$\beta \propto T^L, L=1 \text{ (p-wave)}$$



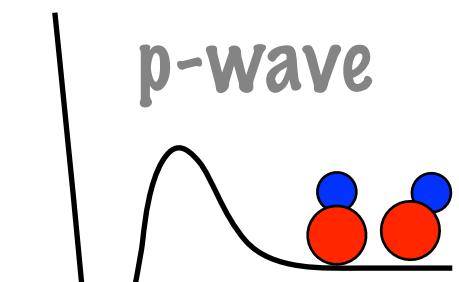
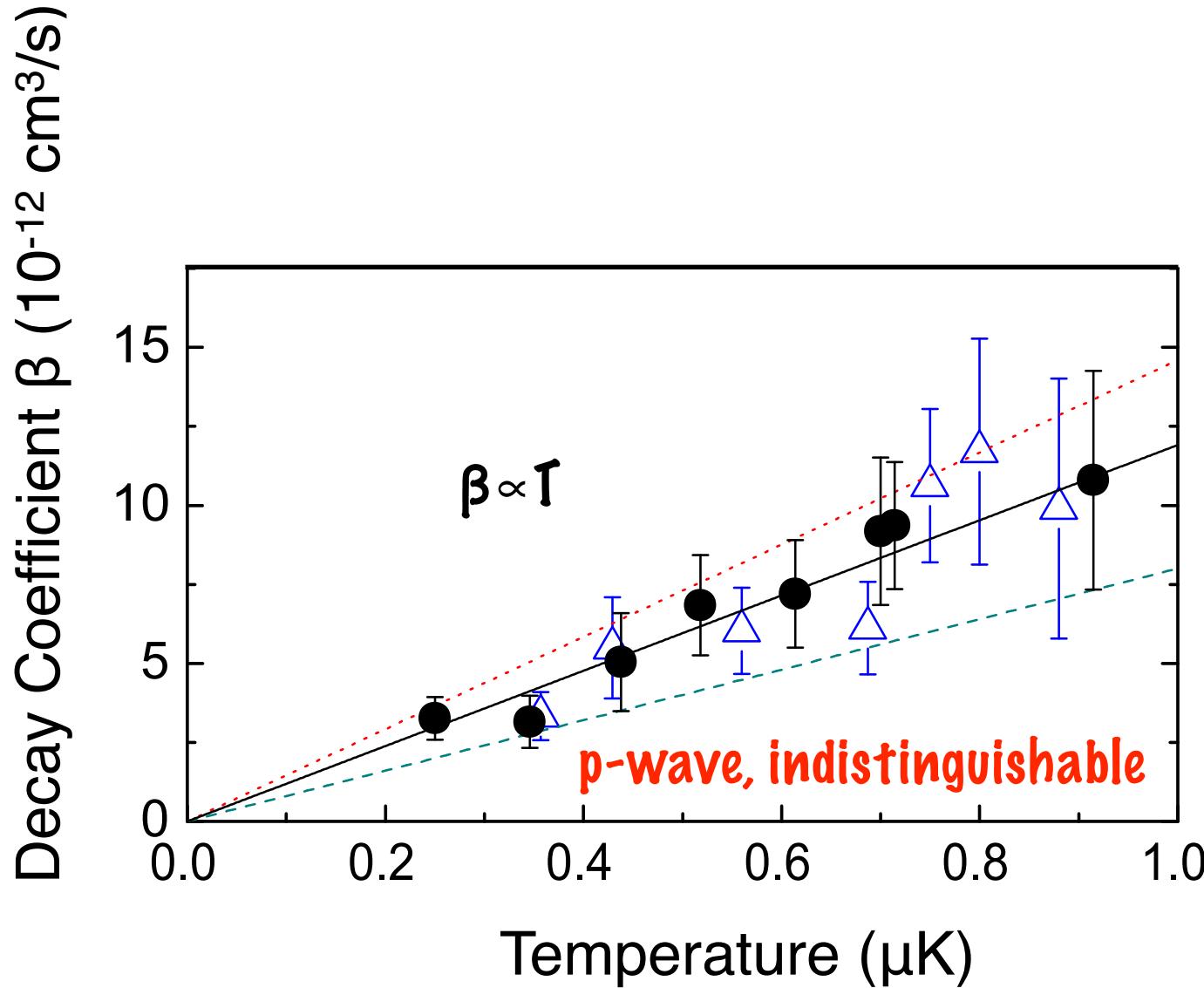
Ultracold Fermi Gas of Molecules

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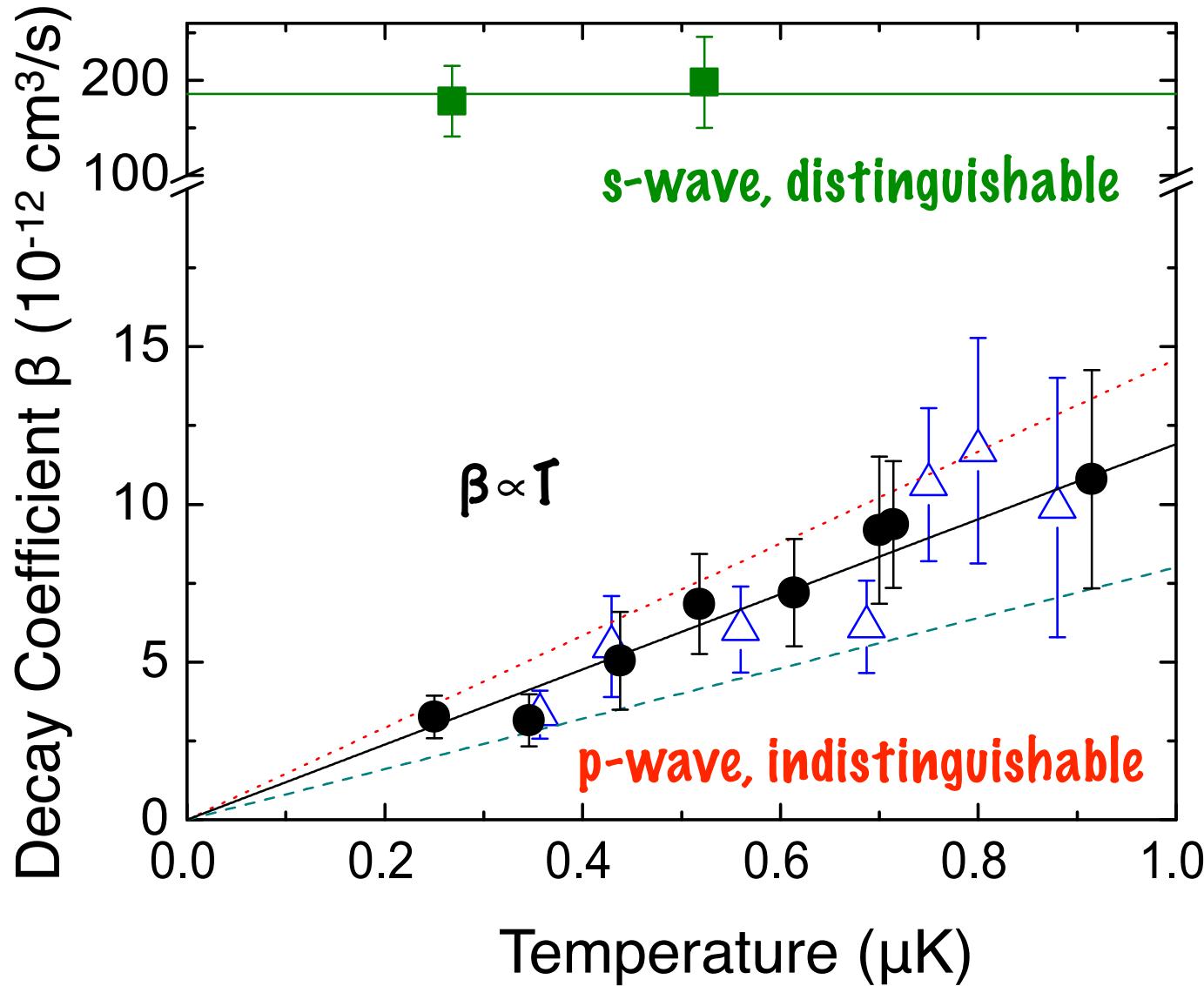
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Molecular Inelastic Collisions - Wigner Threshold Law



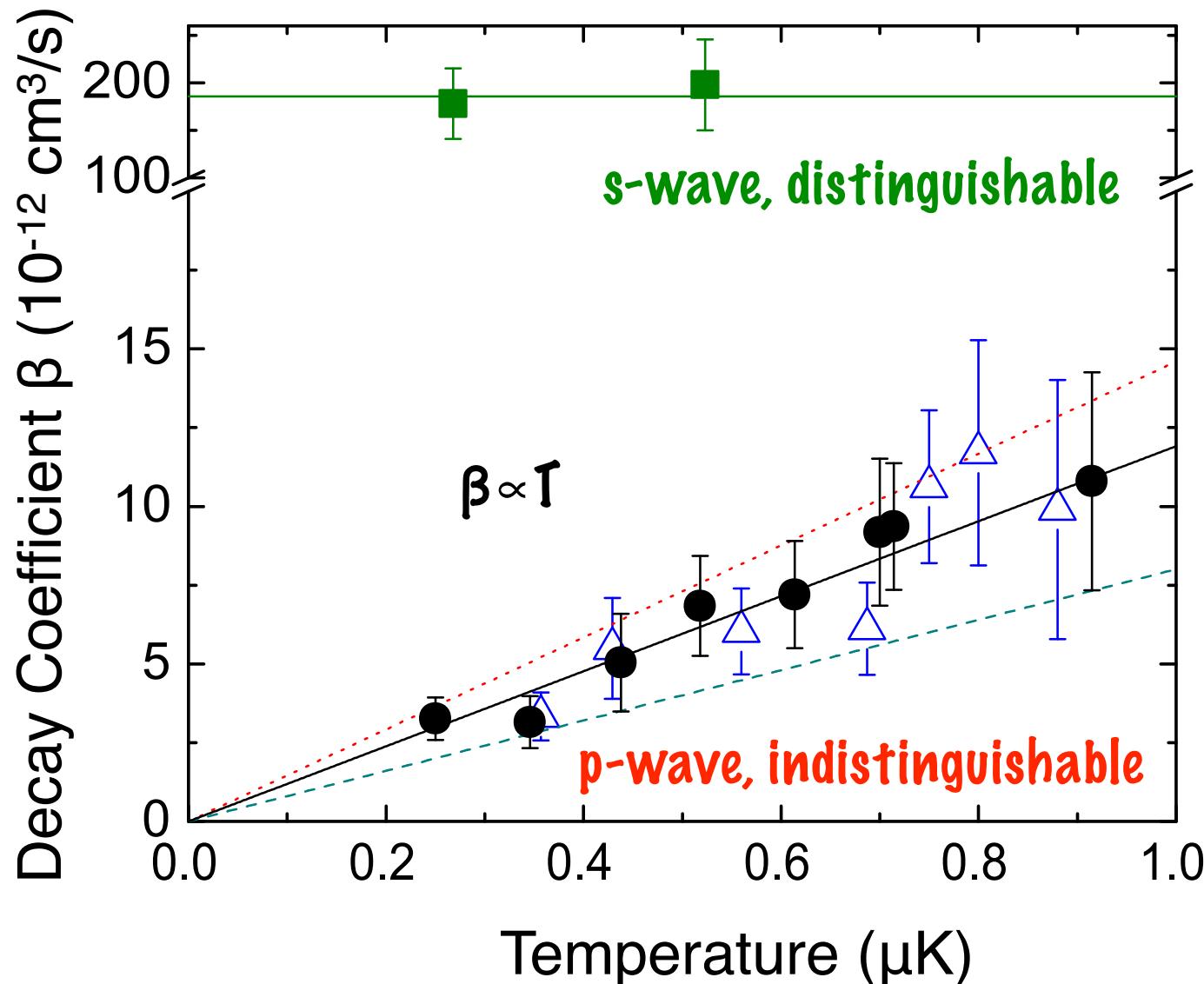
Molecular Inelastic Collisions - Wigner Threshold Law



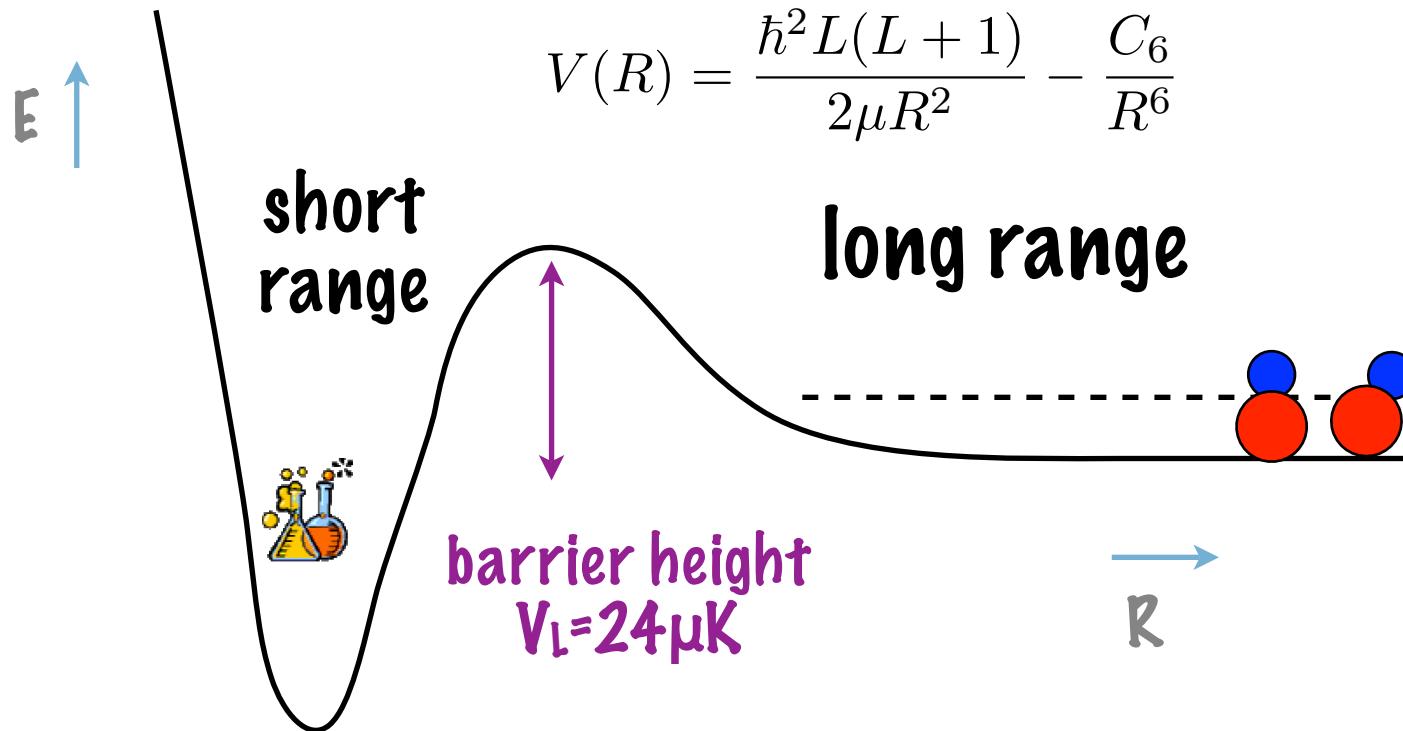
Molecular Inelastic Collisions

- Wigner Threshold Law

Quantum Statistics Controlled Reaction Rates

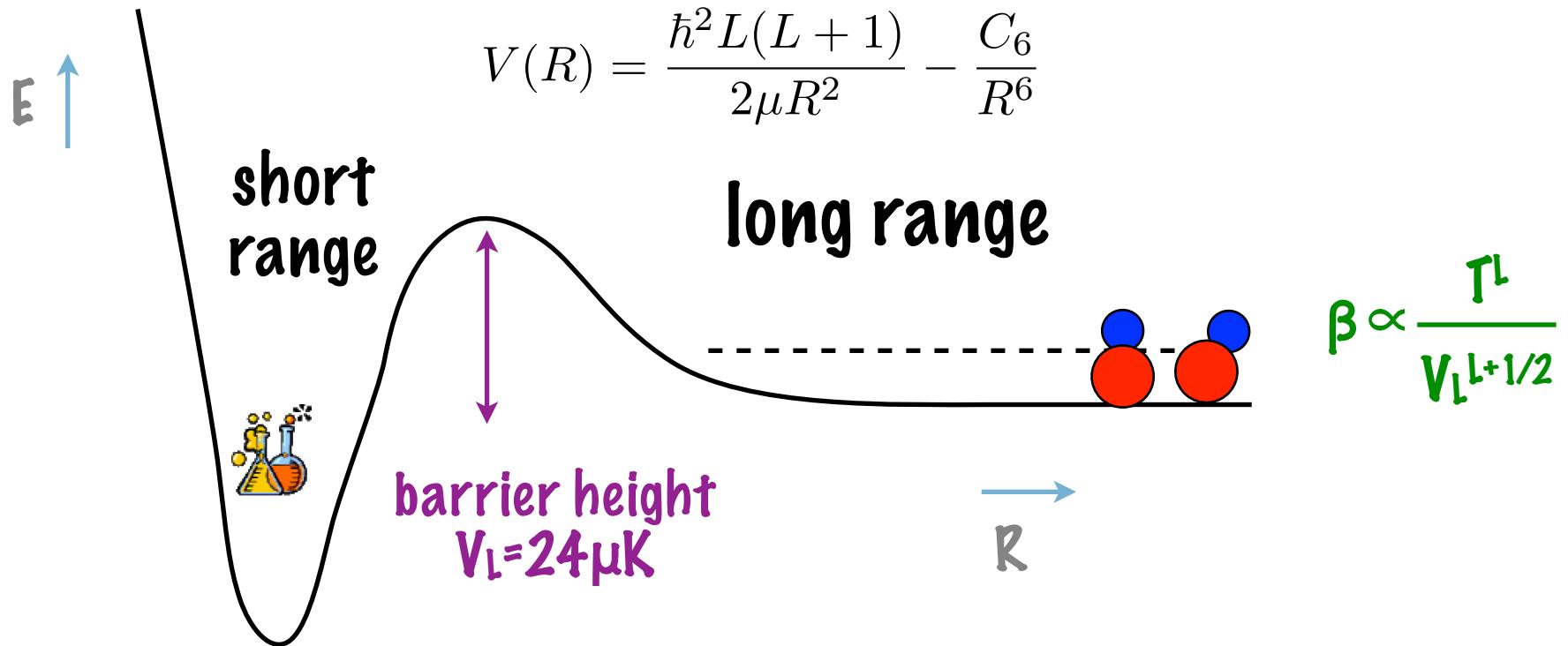


“Quantum Threshold” Model



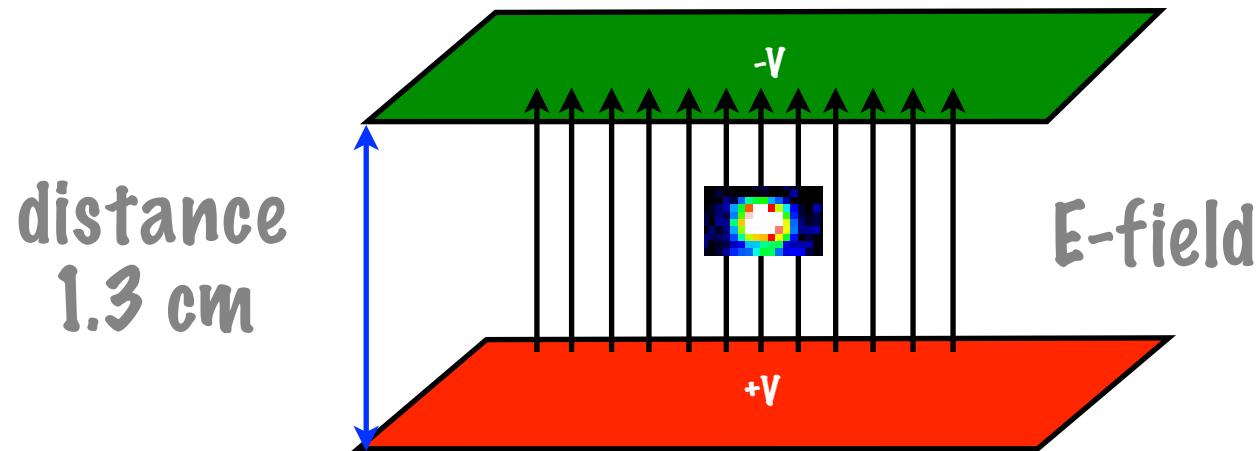
- * Short Range: assume unit probability for loss
- * Long Range: small probability for tunneling through the barrier (rate can be estimated by the barrier height and collisional energy)

“Quantum Threshold” Model

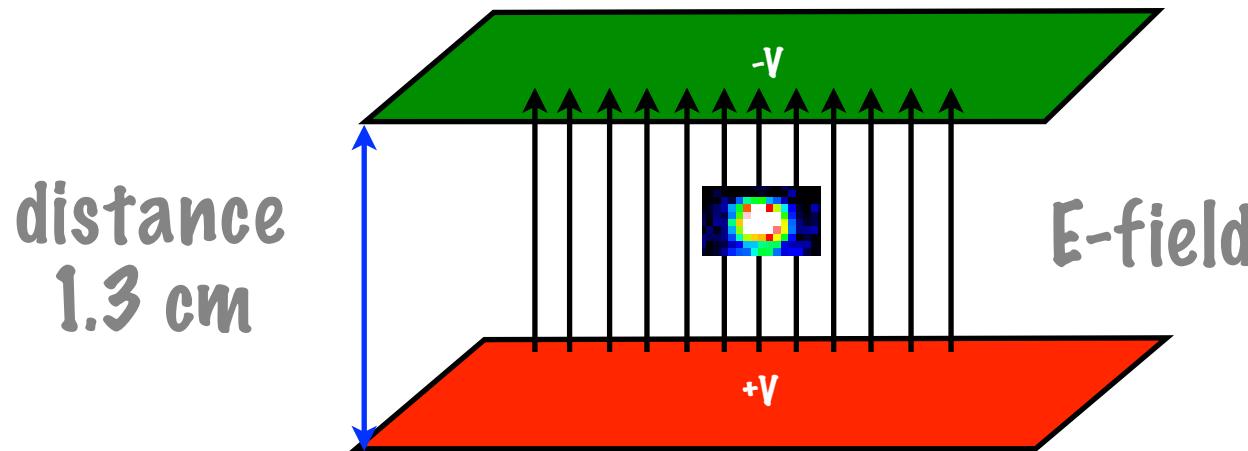


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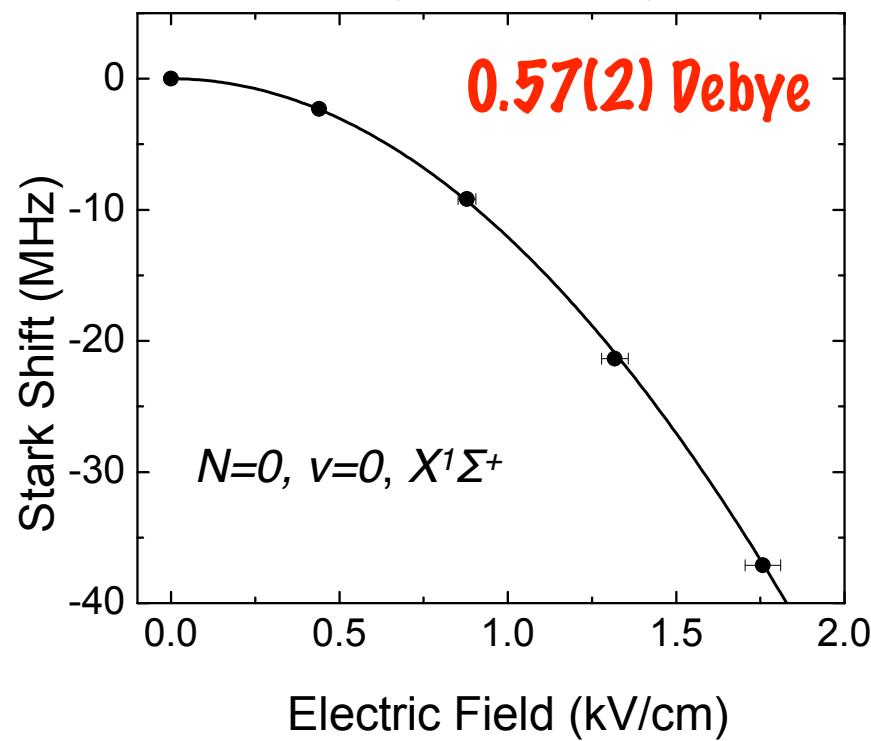
Turn on Electric Field



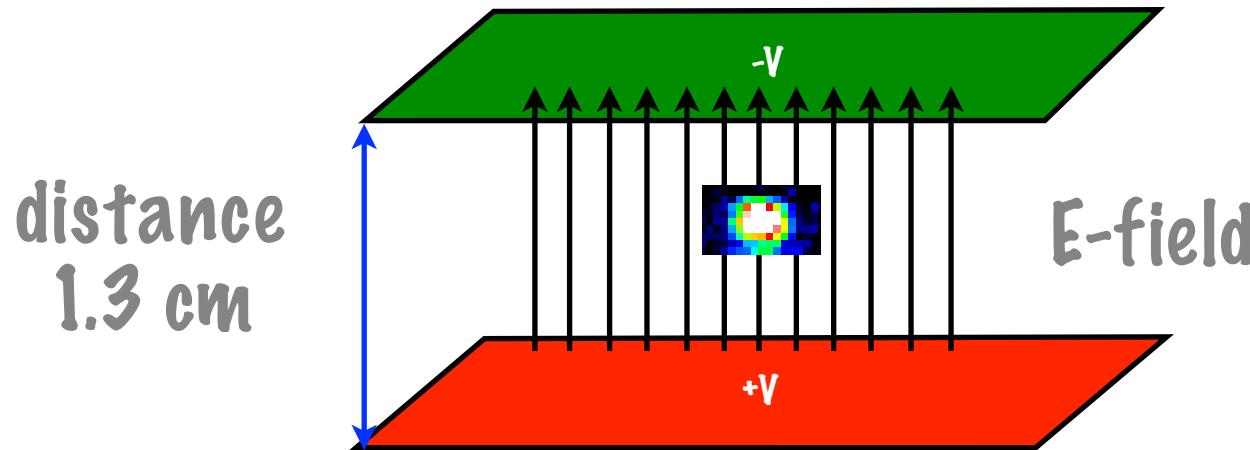
Turn on Electric Field



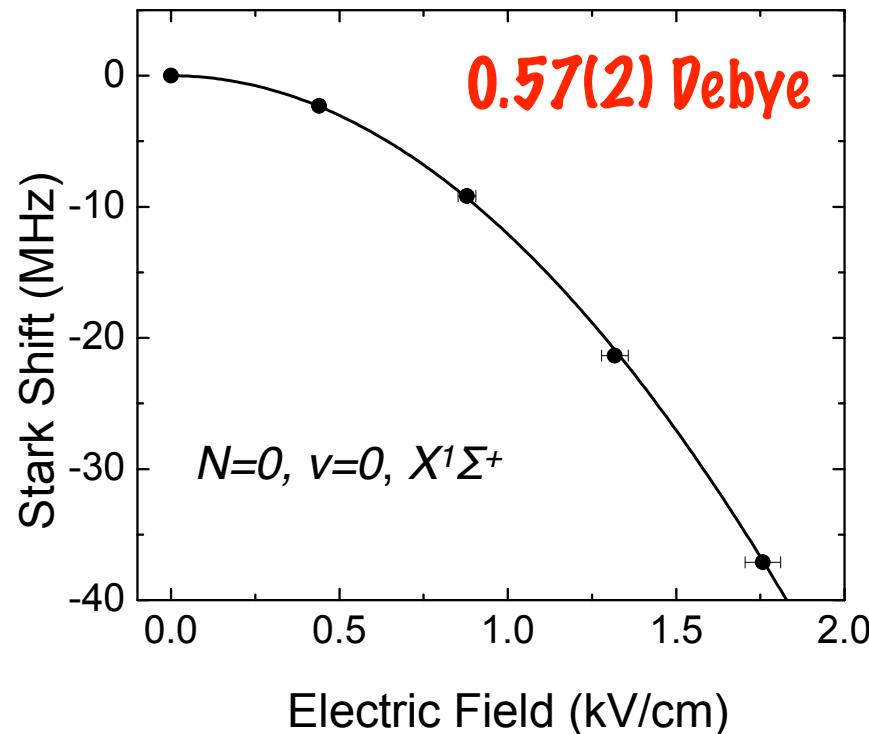
Stark Spectroscopy



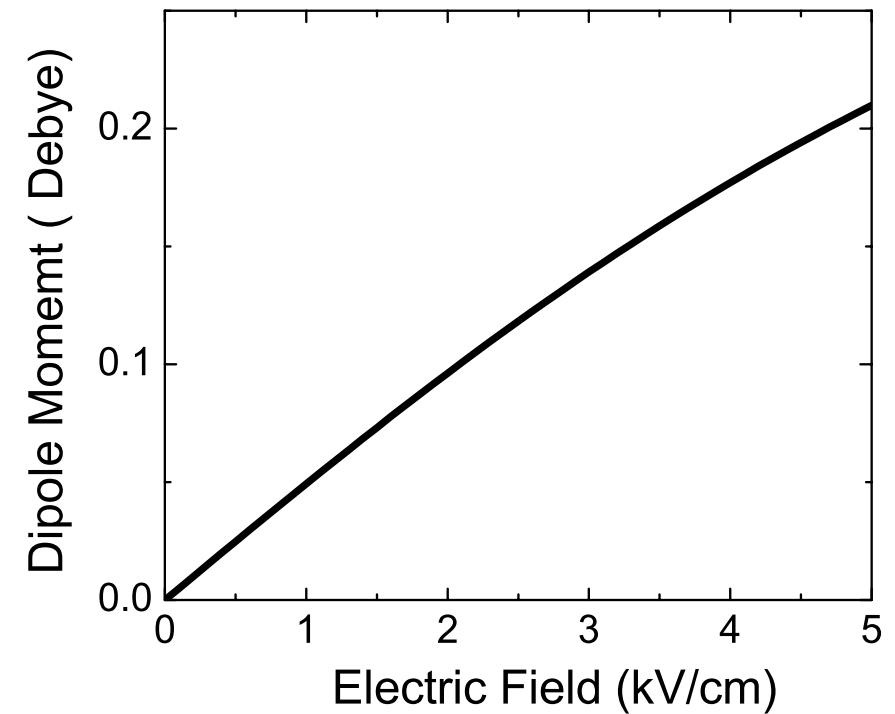
Turn on Electric Field



Stark Spectroscopy



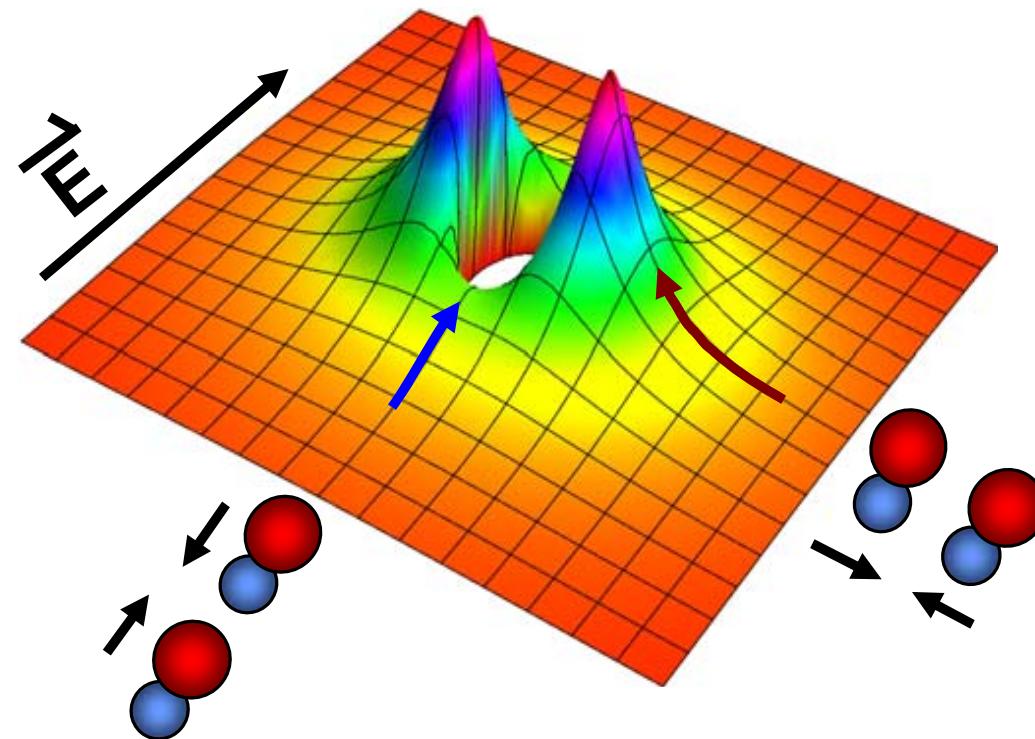
Tuning dipole moment with an E-field



Anisotropy of P-wave Barrier Heights

Long-range potential

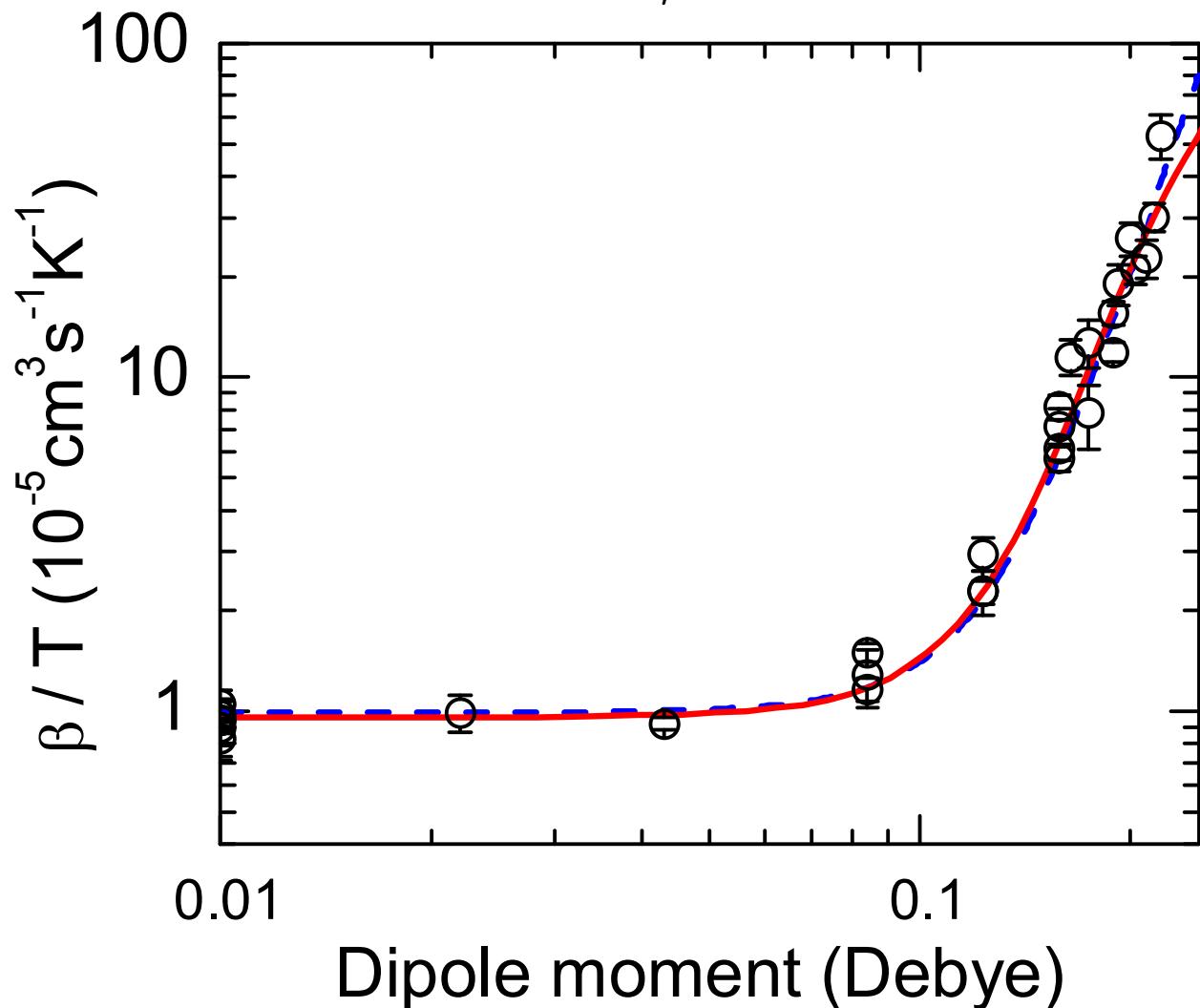
$$V(R) = \frac{\hbar^2 L(L+1)}{2\mu R^2} - \frac{C_6}{R^6} - \frac{C_3}{R^3}$$



collisions in 3D will effectively average over the different channels

Inelastic Dipolar Collisions

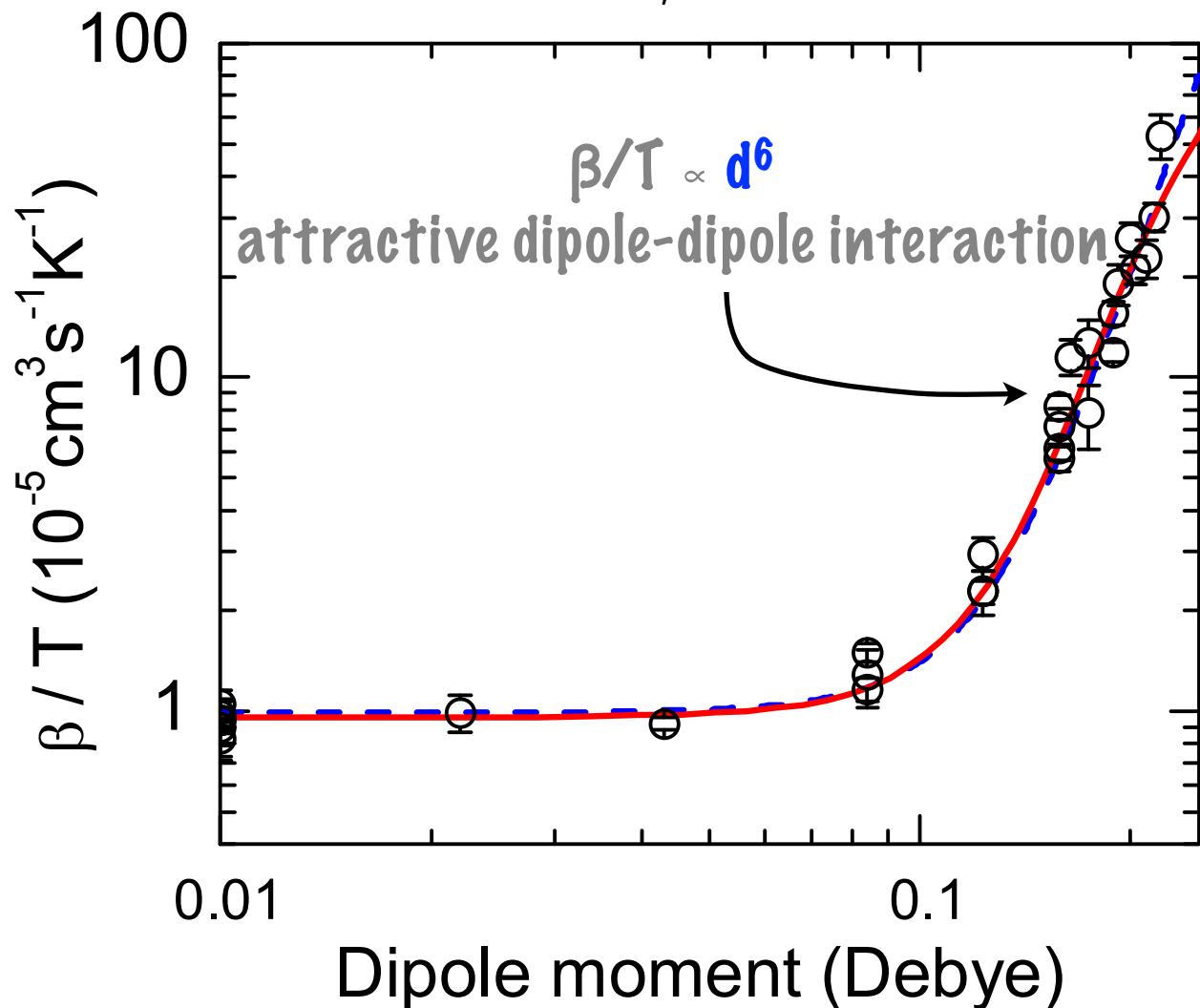
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Theory by
Quéméner
and Bohn

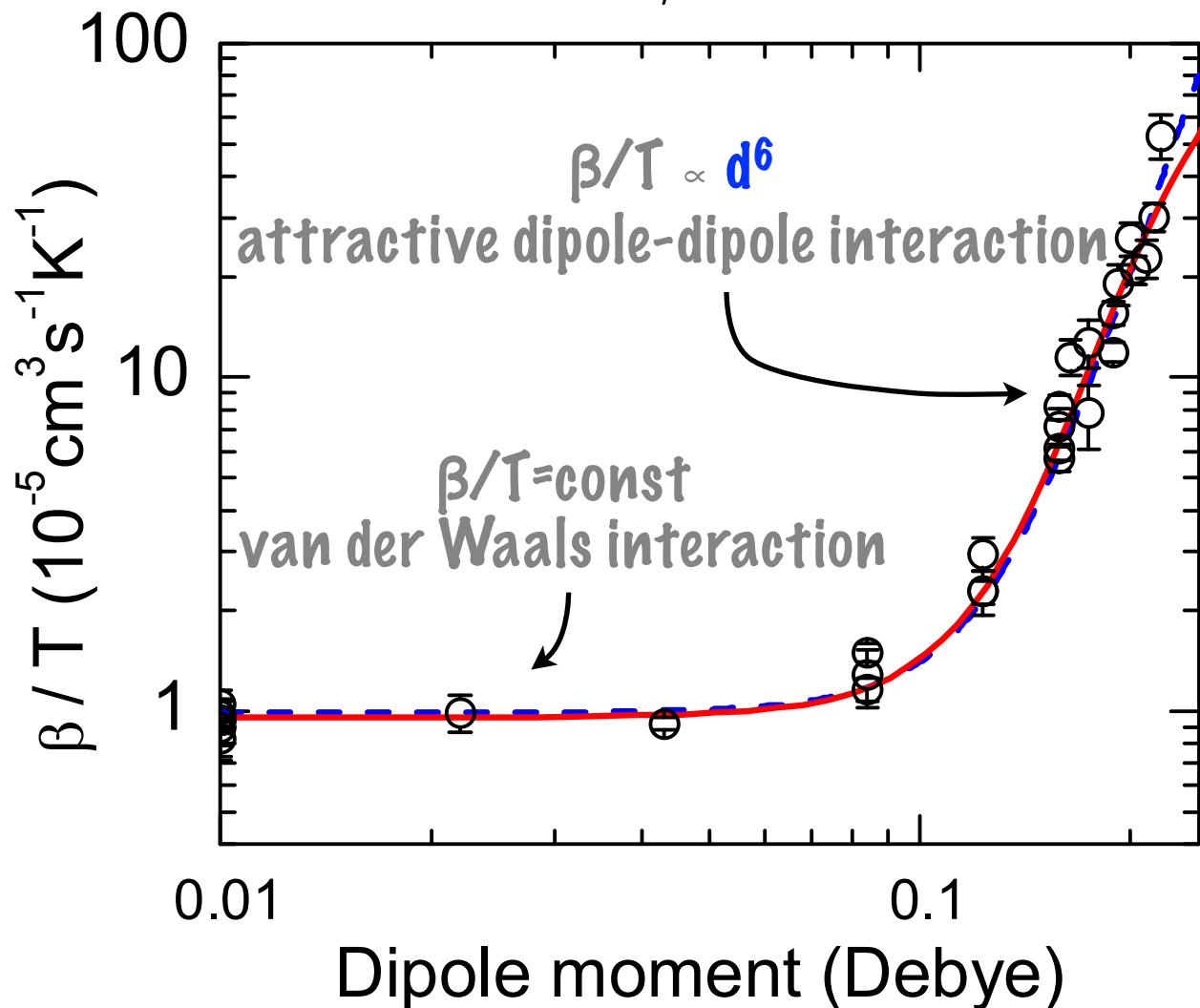
Inelastic Dipolar Collisions

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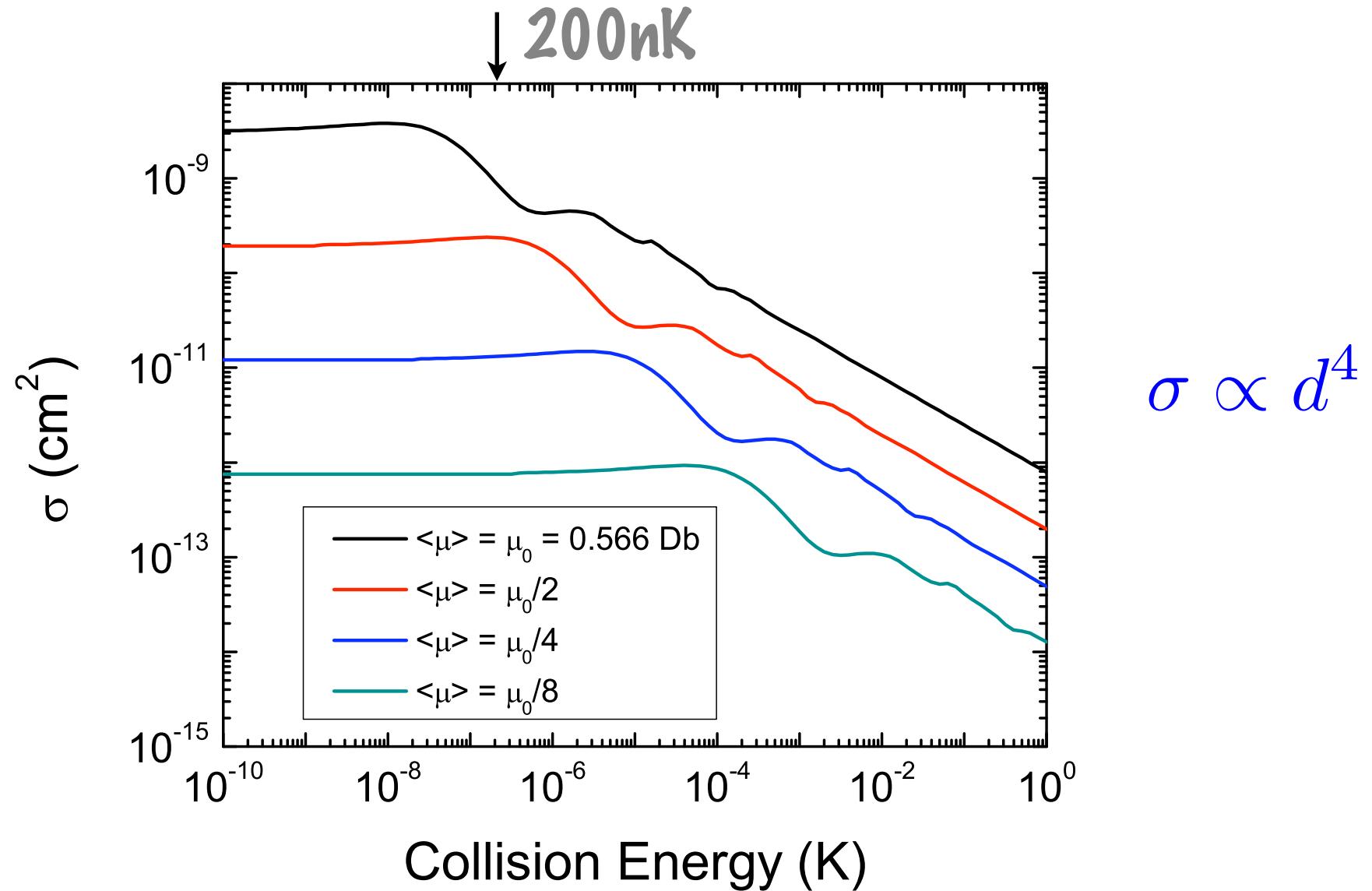
Inelastic Dipolar Collisions

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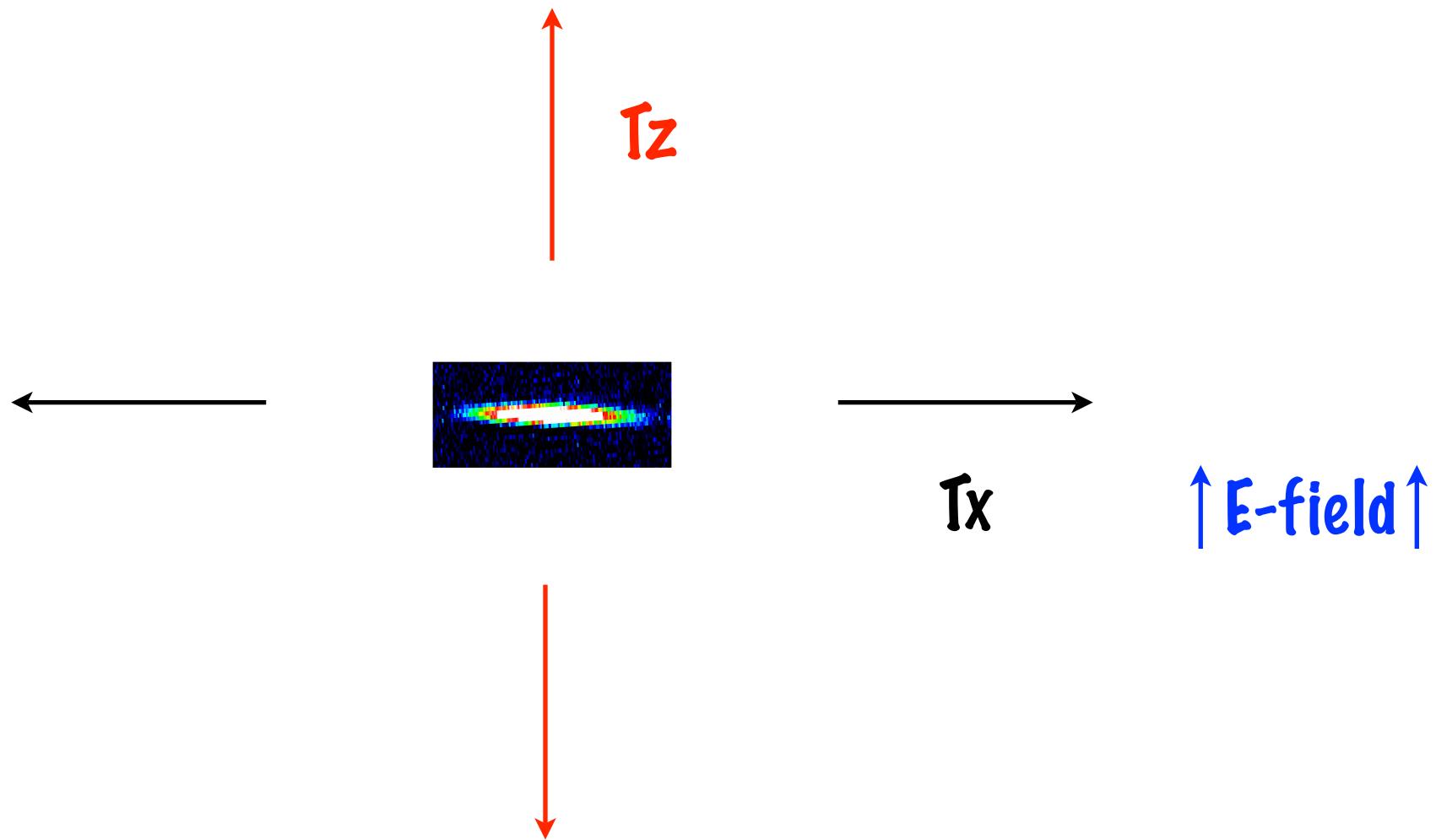


what about elastic collisions?

indistinguishable Fermionic KRb elastic collisions

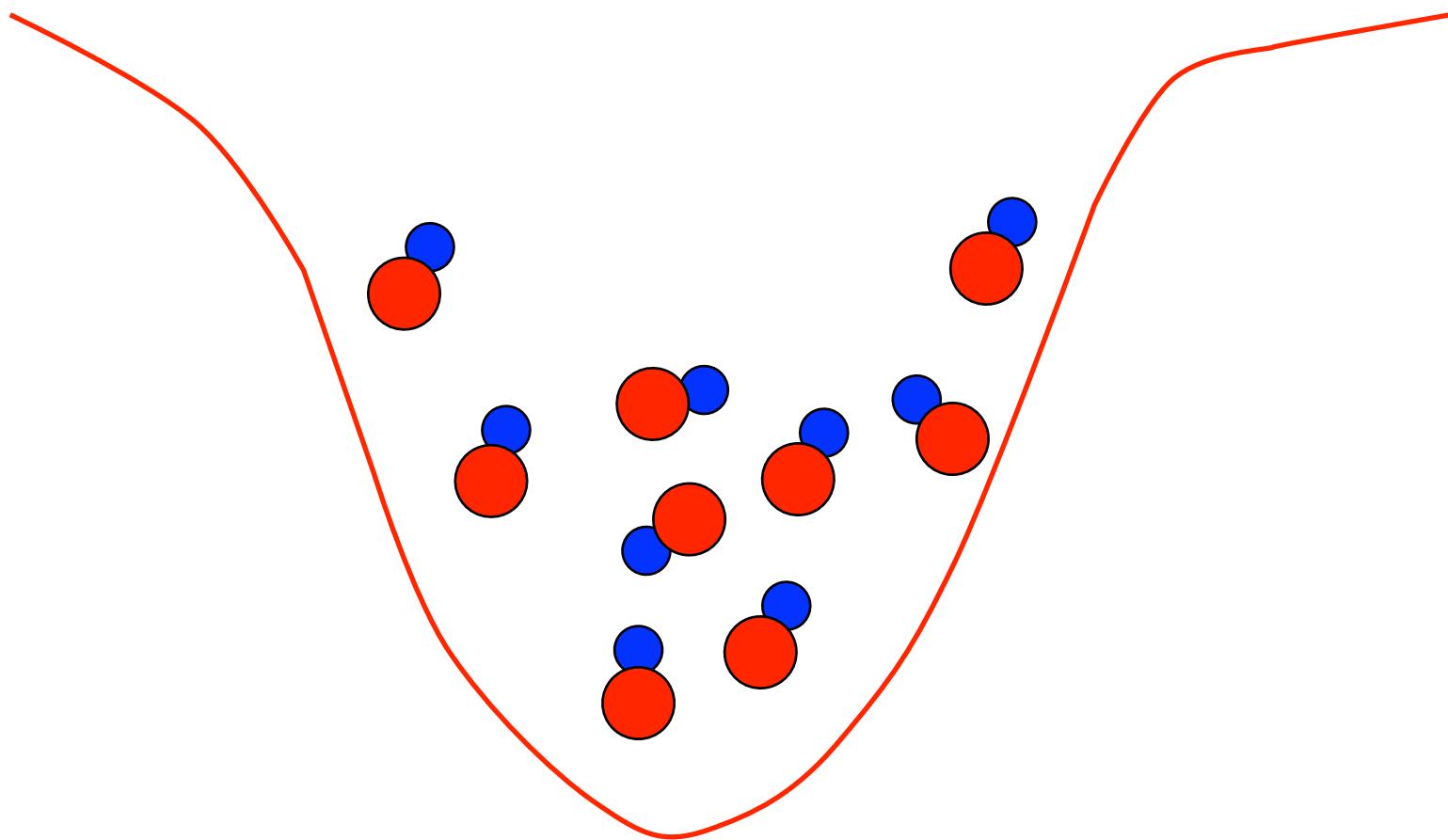


cross-dimension rethermalization



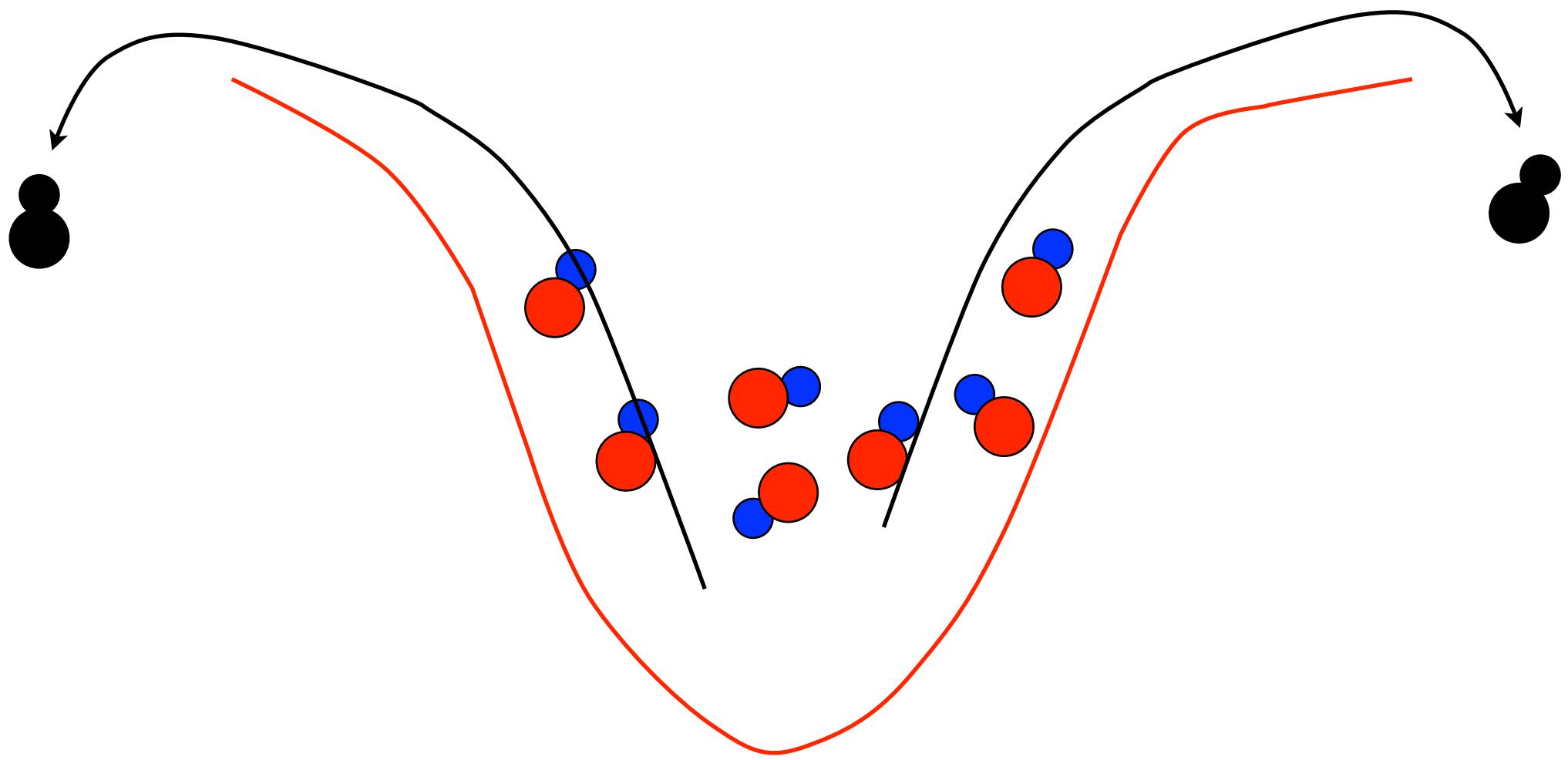
Heating from Inelastic Collisions

$$\dot{n} = -\beta n^2$$



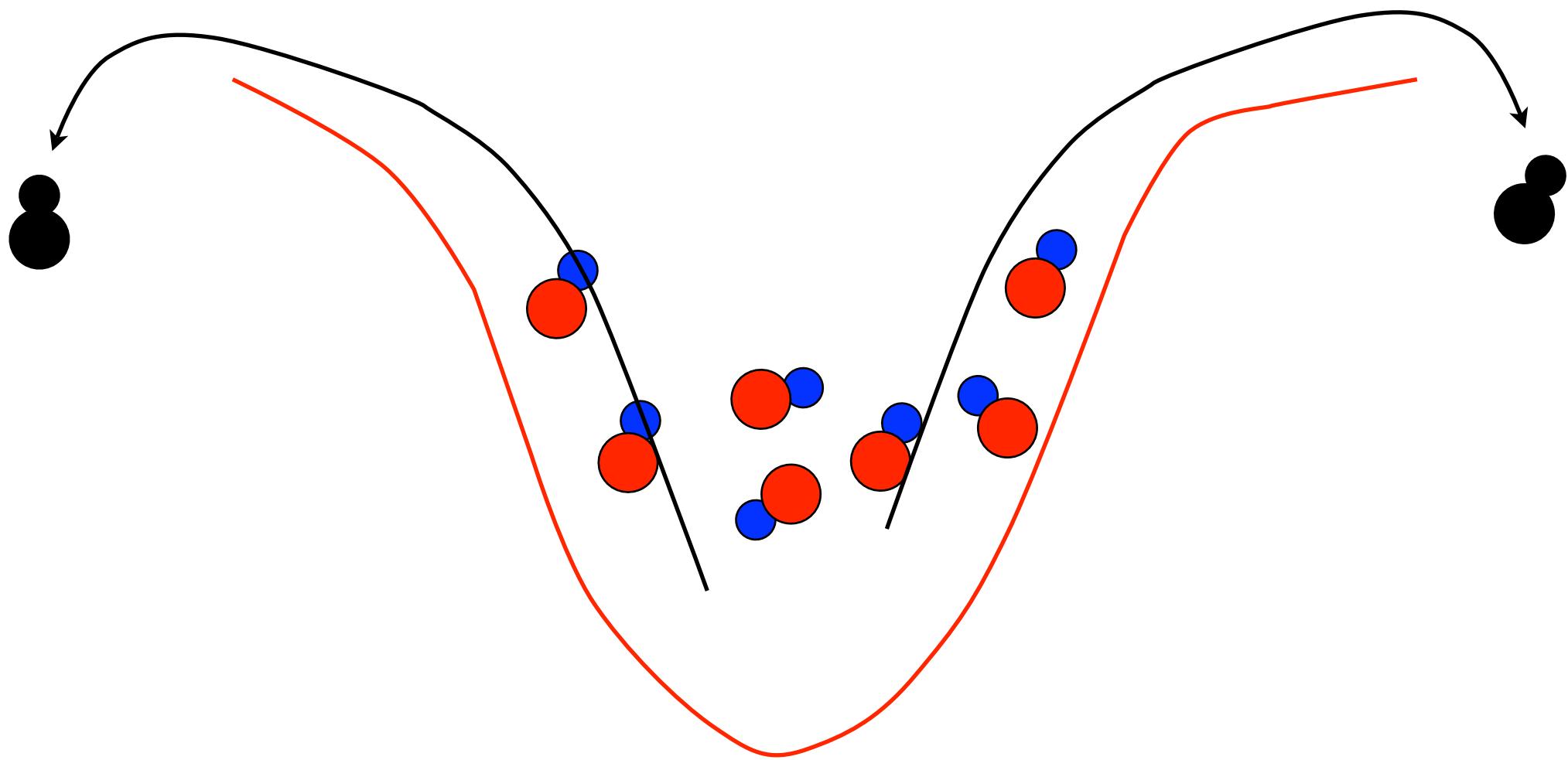
Heating from Inelastic Collisions

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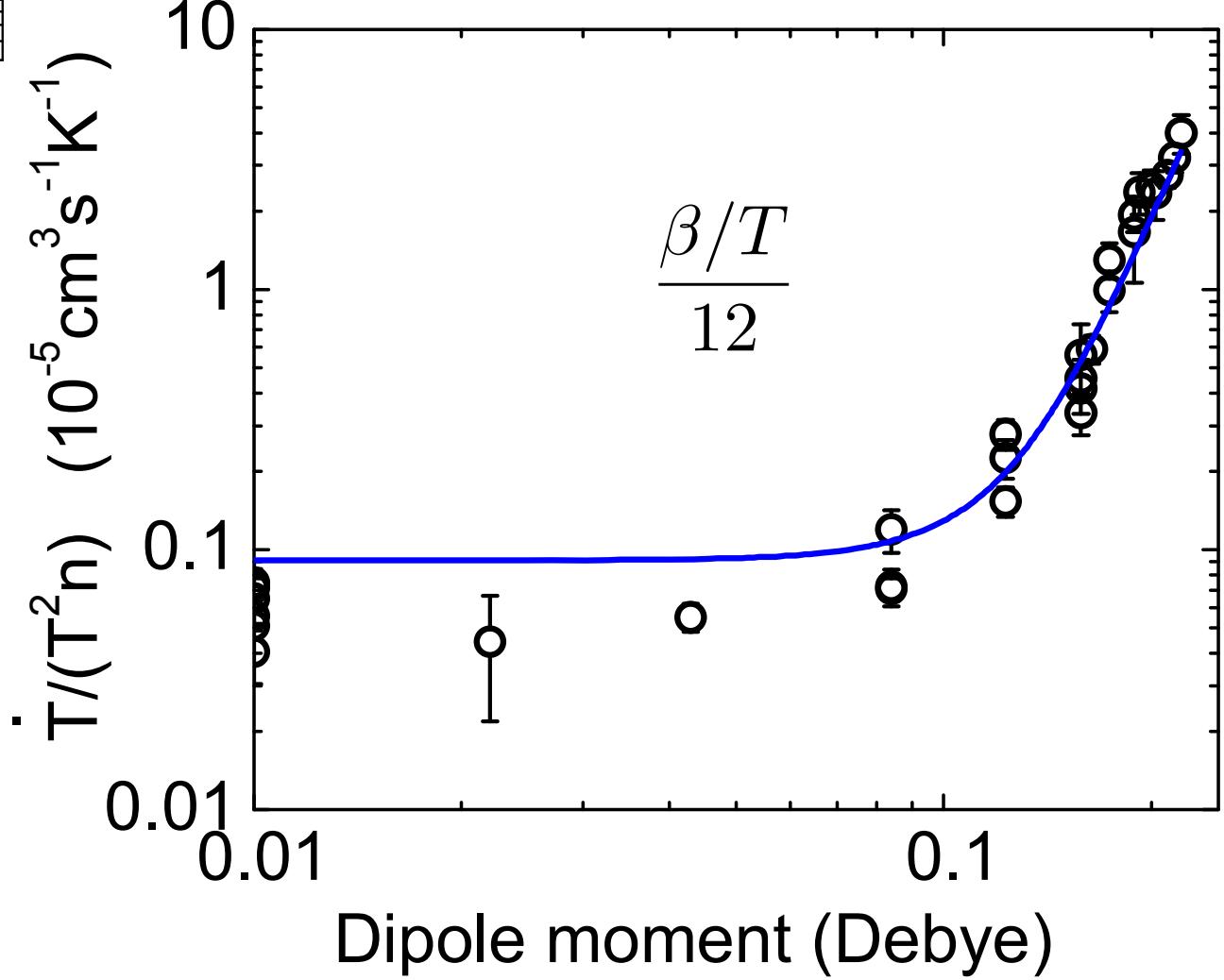
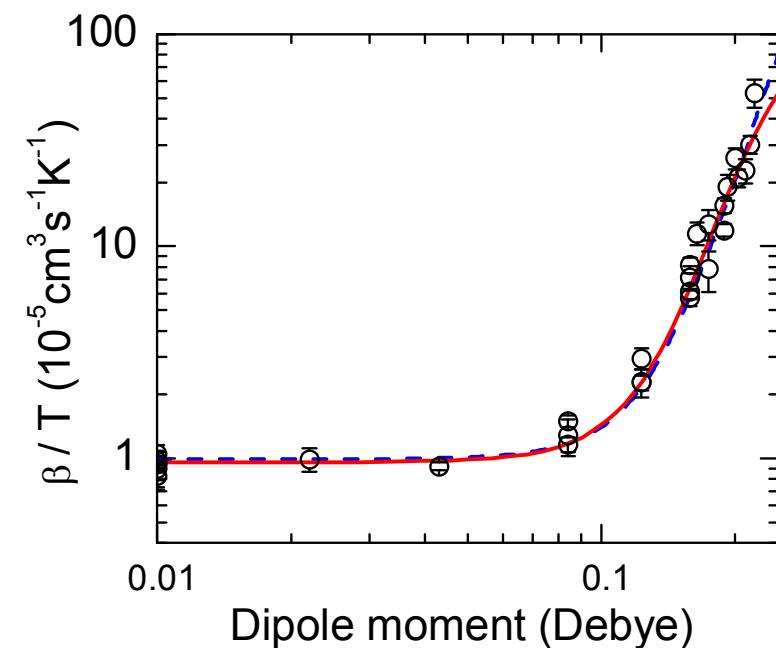
Heating from Inelastic Collisions

$$\dot{n} = -\beta n^2 = -K \cdot T \cdot n^2$$



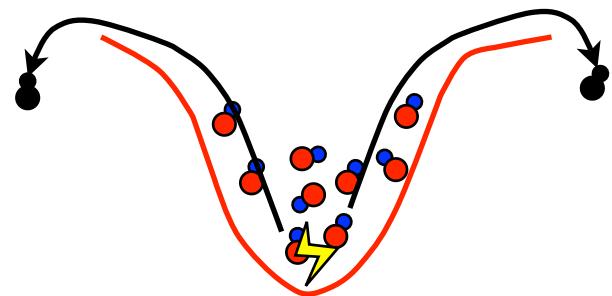
Heating

$$T = \frac{(2T_x + T_z)}{3}$$



(Anisotropic) Heating or Cooling?

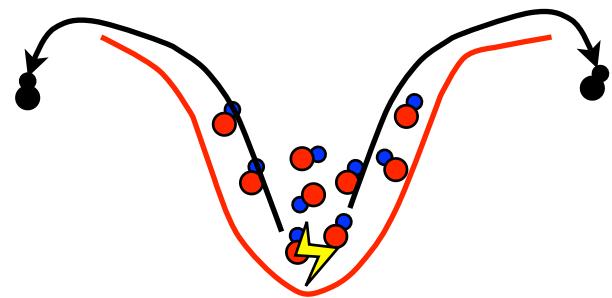
$$\Delta E_z^{m_L=0} = 2k_b T_z - k_b T_z - \frac{\mu}{2} \frac{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^4 dv_z}{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^2 dv_z}$$



(Anisotropic) Heating or Cooling?

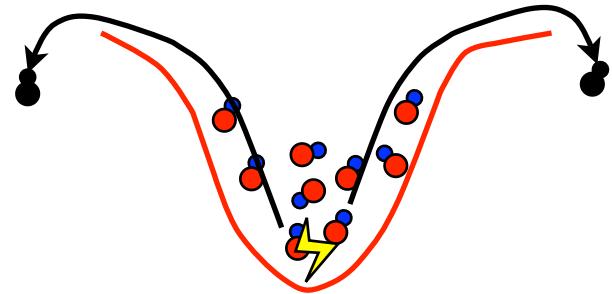
$$\Delta E_z^{m_L=0} = \underbrace{2k_b T_z}_{\text{average energy for a pair}} - k_b T_z - \frac{\mu}{2} \frac{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^4 dv_z}{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^2 dv_z}$$

average energy
for a pair



(Anisotropic) Heating or Cooling?

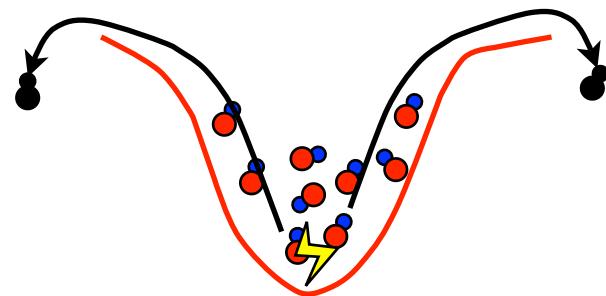
$$\Delta E_z^{m_L=0} = \underbrace{2k_b T_z}_{\text{average energy for a pair}} - \underbrace{k_b T_z}_{\text{average c.m. energy}} - \frac{\mu}{2} \frac{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^4 dv_z}{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^2 dv_z}$$



(Anisotropic) Heating or Cooling?

averaging for relative kinetic energy of a pair

$$\Delta E_z^{m_L=0} = \underbrace{2k_b T_z}_{\text{average energy}} - \underbrace{k_b T_z}_{\text{average c.m. energy}} - \frac{\mu}{2} \underbrace{\frac{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^4 dv_z}{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^2 dv_z}}_{\text{MB distribution,}}$$

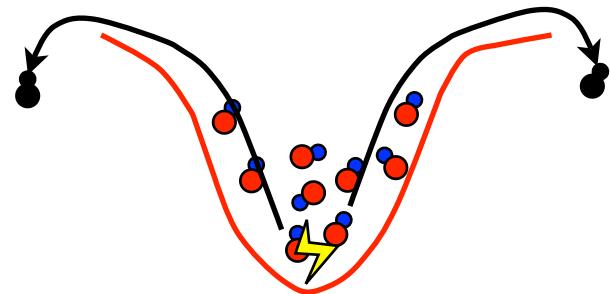


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cooling!!



(Anisotropic) Heating or Cooling?

averaging for relative kinetic energy of a pair

$$\Delta E_z^{m_L=0} = \underbrace{2k_b T_z}_{\text{average energy}} - \underbrace{k_b T_z}_{\text{average c.m. energy}} - \frac{\mu}{2} \underbrace{\frac{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^4 dv_z}{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^2 dv_z}}_{\text{MB distribution,}} = 2k_b T_z - \frac{5}{2} k_b T_z = -\frac{1}{2} k_b T_z$$

cooling!!

$$\Delta E_x^0 = \frac{1}{2} k_b T_x$$

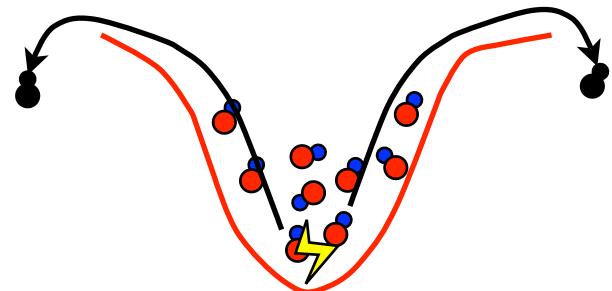
heating!!

$$\Delta E_z^1 = \frac{1}{2} k_b T_z$$

heating!!

$$\Delta E_x^1 = 0$$

no change



(Anisotropic) Heating or Cooling?

averaging for relative kinetic energy of a pair

$$\Delta E_z^{m_L=0} = \underbrace{2k_b T_z}_{\text{average energy for a pair}} - \underbrace{k_b T_z}_{\text{average c.m. energy}} - \frac{\mu}{2} \underbrace{\frac{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^4 dv_z}{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^2 dv_z}}_{\text{MB distribution,}} = 2k_b T_z - \frac{5}{2} k_b T_z = -\frac{1}{2} k_b T_z$$

cooling!!

$$\Delta E_x^0 = \frac{1}{2} k_b T_x$$

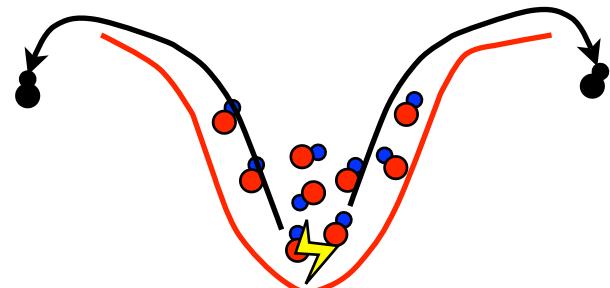
heating!!

$$\Delta E_z^1 = \frac{1}{2} k_b T_z$$

heating!!

$$\Delta E_x^1 = 0$$

no change
Heating Rate



$$\frac{dT_z}{dt} = \frac{n}{2} \left(\frac{-1}{2} T_z \beta^0 + \frac{1}{2} T_z \beta^1 \right)$$

$$\frac{dT_x}{dt} = \frac{n}{2} \left(\frac{1}{2} T_x \beta^0 \right)$$

(Anisotropic) Heating or Cooling?

averaging for relative kinetic energy of a pair

$$\Delta E_z^{m_L=0} = \underbrace{2k_b T_z}_{\text{average energy for a pair}} - \underbrace{k_b T_z}_{\text{average c.m. energy}} - \frac{\mu}{2} \underbrace{\frac{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^4 dv_z}{\int_{-\infty}^{\infty} f(v_z, v_r) v_z^2 dv_z}}_{\text{MB distribution,}} = 2k_b T_z - \frac{5}{2} k_b T_z = -\frac{1}{2} k_b T_z$$

cooling!!

$$\Delta E_x^0 = \frac{1}{2} k_b T_x$$

heating!!

$$\Delta E_z^1 = \frac{1}{2} k_b T_z$$

heating!!

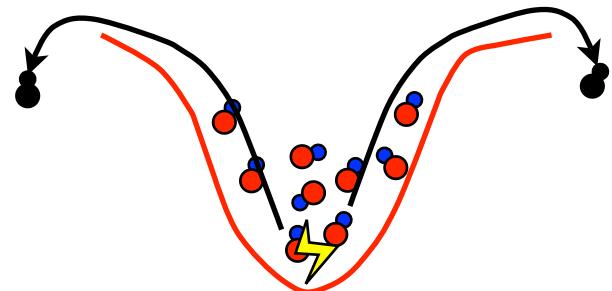
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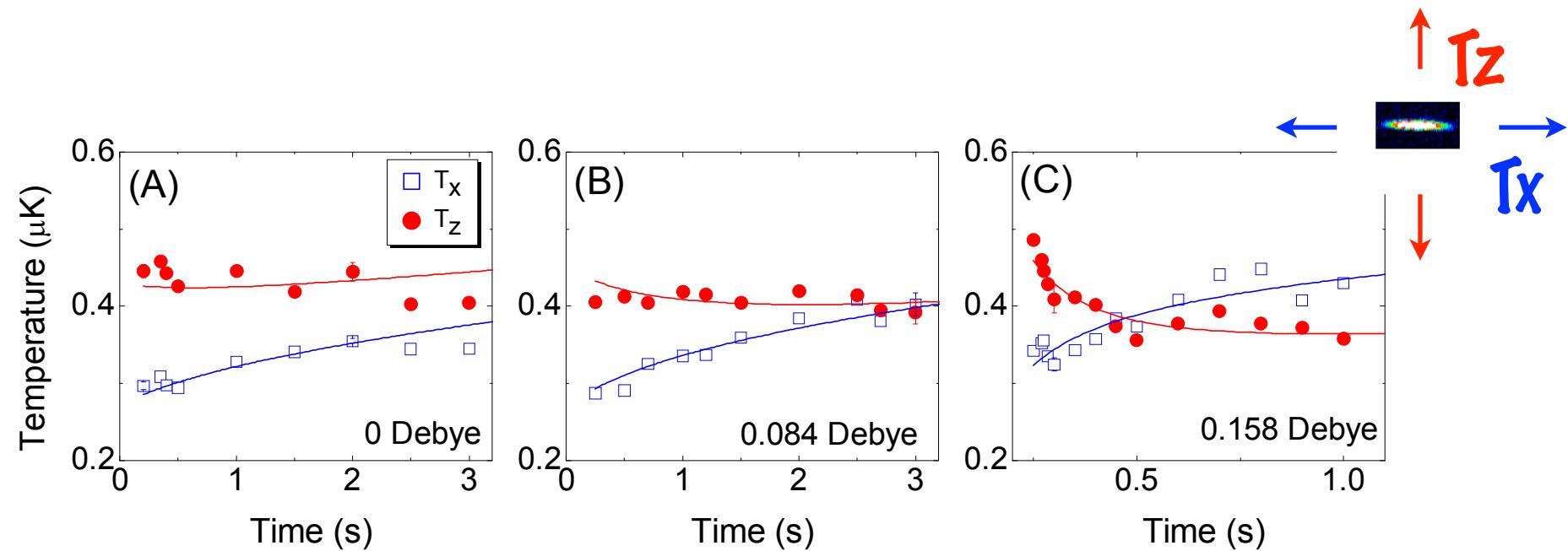
Heating Rate

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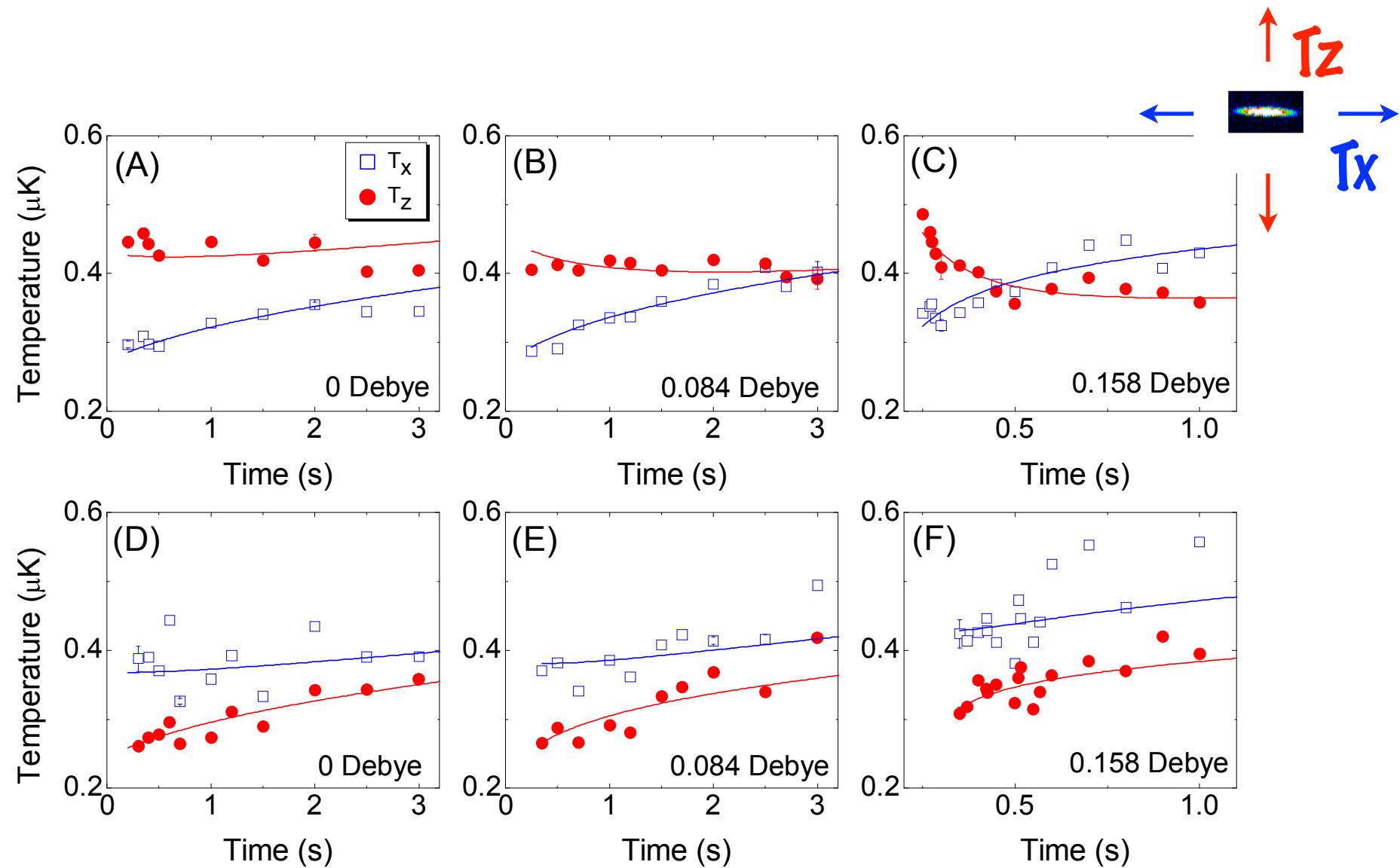
$$\frac{dT_x}{dt} = \frac{n}{2} \left(\frac{1}{2} T_x \beta^0 \right)$$



Evidence of Anisotropic Dipolar Collisions



Evidence of Anisotropic Dipolar Collisions



Observed anisotropic inelastic collisions,
but the heating/cooling effects eclipse
elastic rethermalization collisions

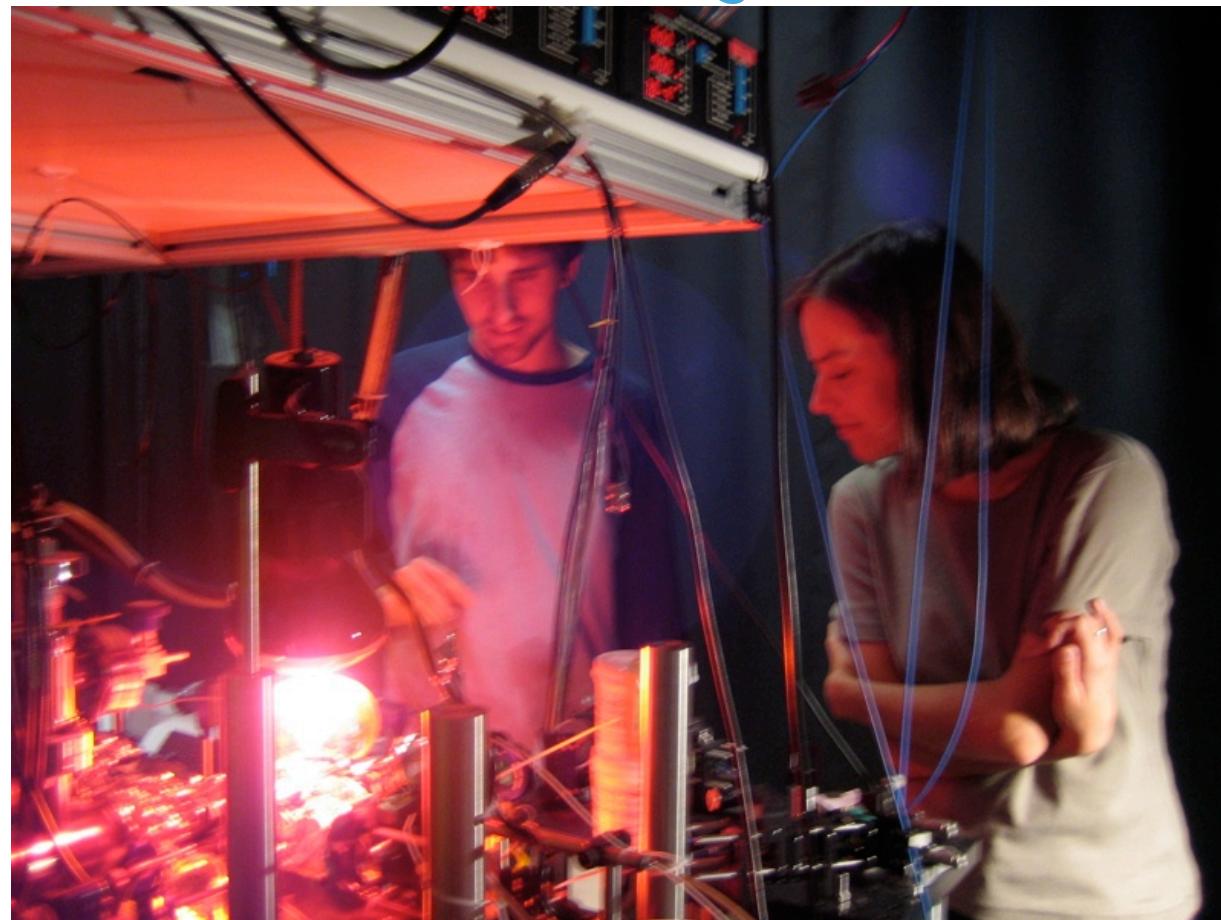
Conclusions

- * >90% efficiency, fewx10⁴ polar molecules at 160nK.
 $T/T_F = 1.4$ (PSD 0.06) Peak density 10¹²/cm³
- * a general scheme for other **bialkali** molecules
(on-going effort worldwide)
- * controlling molecular **hyperfine** states
- * seeing ultracold **chemical reactions** !!
- * exploring **long-range and anisotropic dipolar** collisions

Outlook

- * suppress inelastic collisions in 2D geometry
(current on-going work in JILA)
- * evaporative cooling of molecules
- * study of quantum phases

Acknowledgements



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